

Surface studies of crystals

Bogdan J. Kowalski

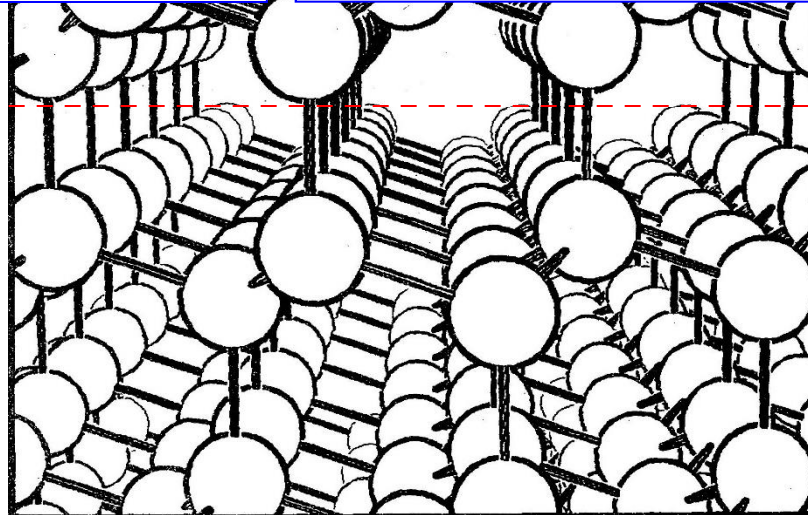
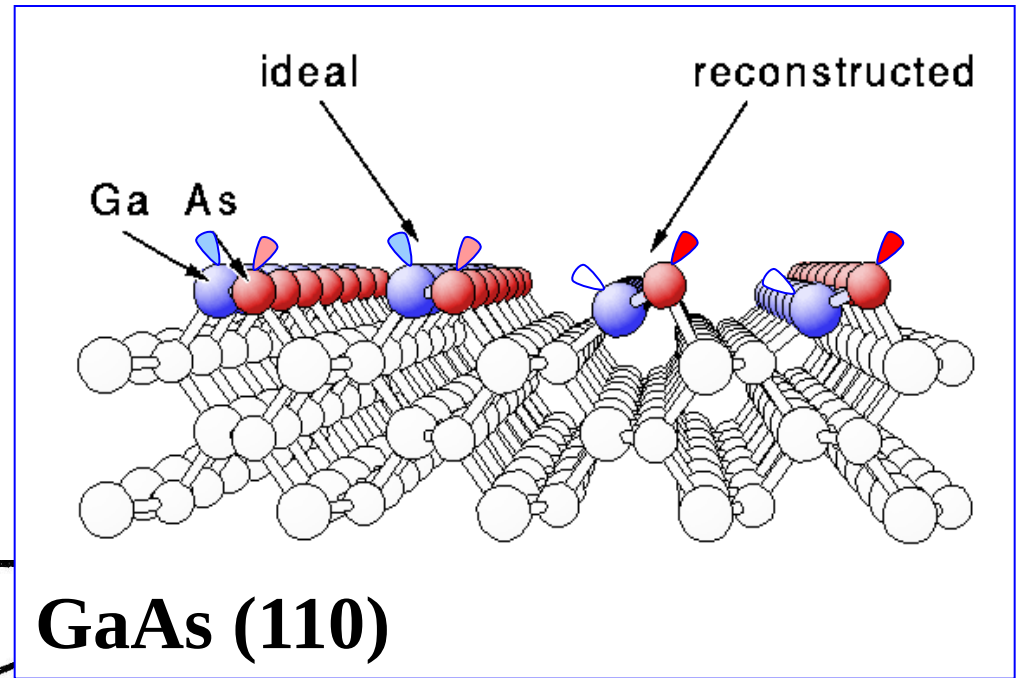
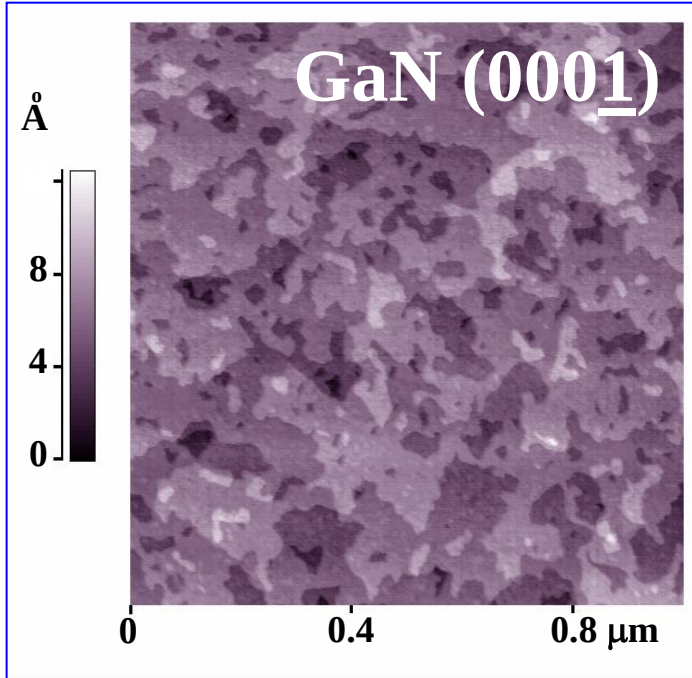
IF PAN

Outline

- **Introduction**
- **Methods:**
 - Electron microscopy**
 - Scanning probe microscopies**
 - Electron spectroscopies**
 - Diffraction methods**
 - Ion techniques**
 - Surface-sensitive optical techniques**
- **Summary, literature**

**What does it
mean
surface?**

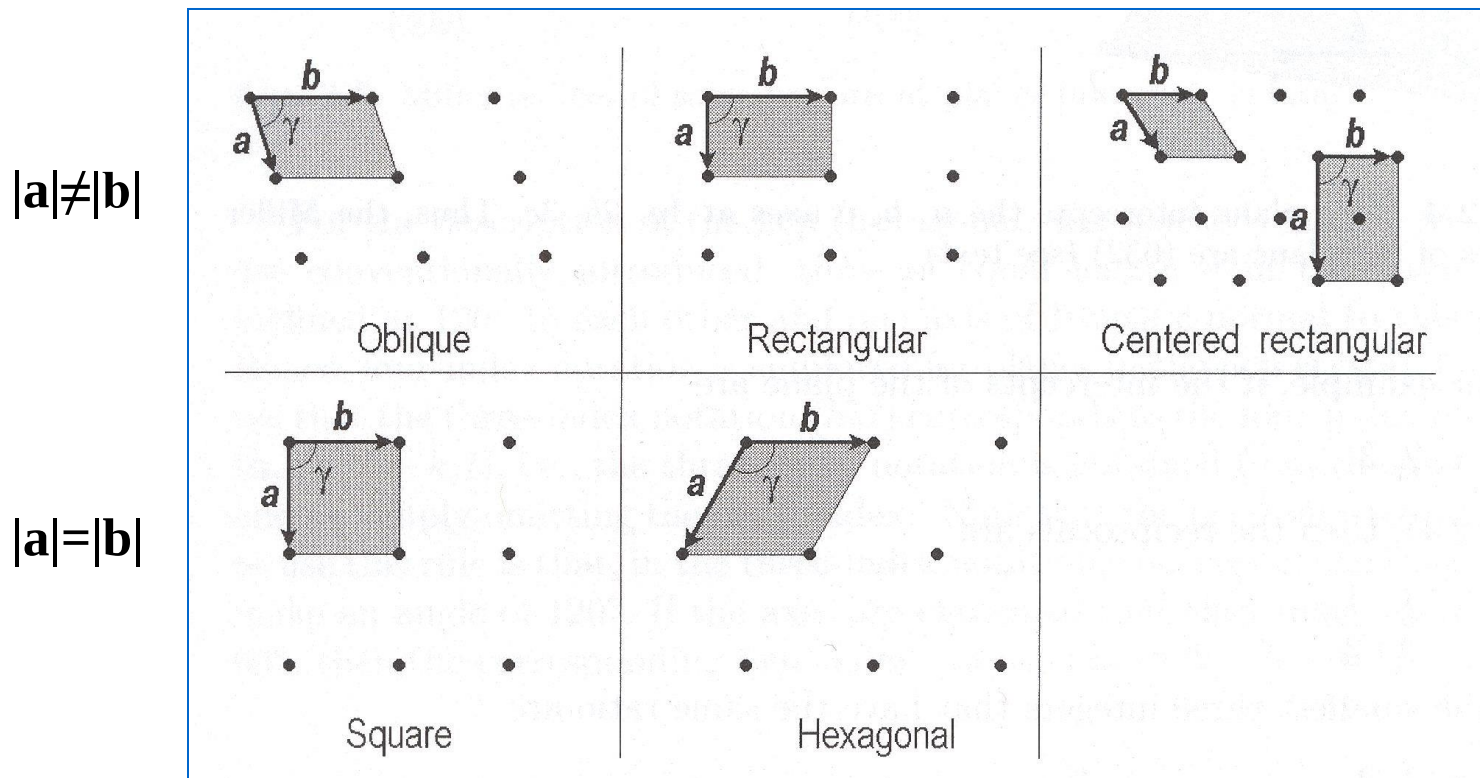
Surface



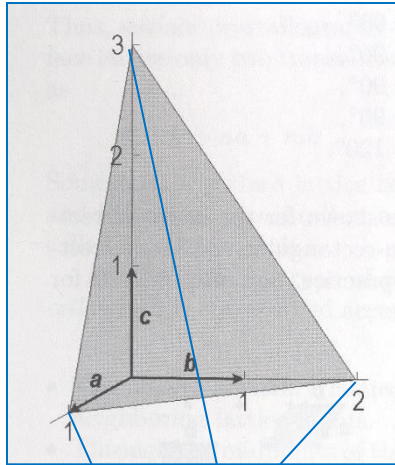
Surface description: Bravais lattices

5 2-dimensional Bravais lattices

(14 3-dimensional Bravais lattices)



Surface description: Miller indices



1, 2, 3

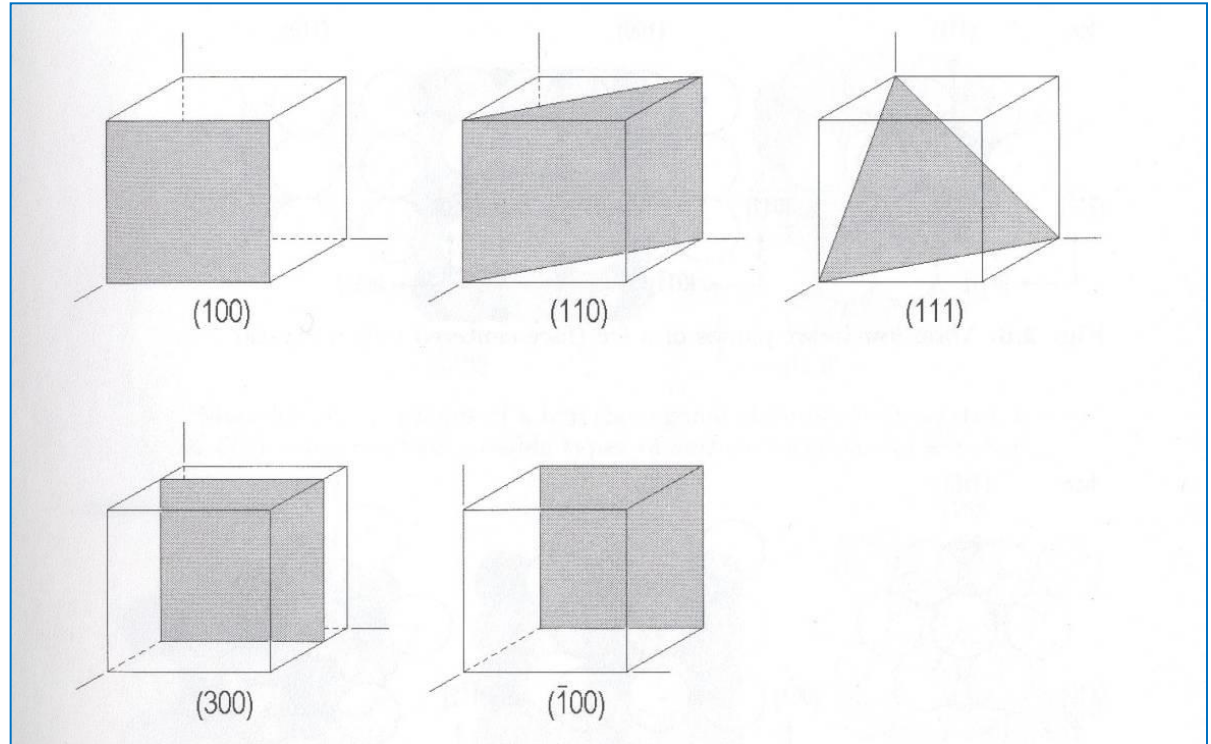


1, 1/2, 1/3



(6, 3, 2) - 1 plane
(h, k, l)

{6, 3, 2} - a set of parallel planes

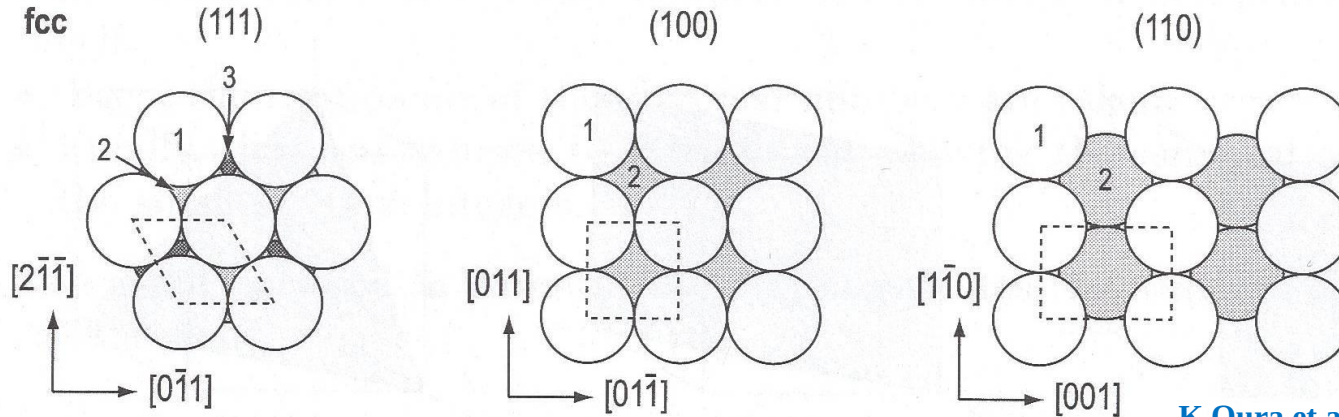


K.Oura et al.
Surface Science. An Introduction

For a hexagonal structure
(h, k, -h-k, l)

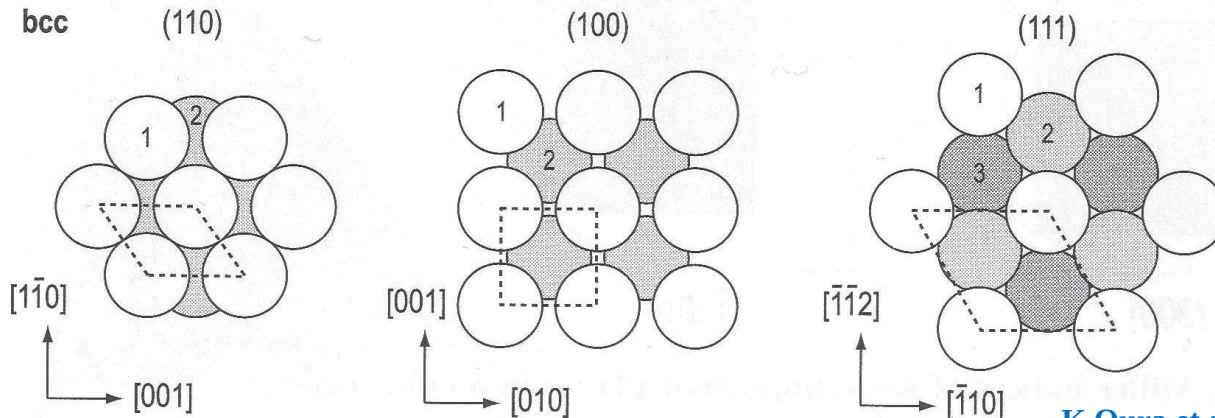
Atomic structure of surfaces - examples

Face-centered cubic crystal



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Body-centered cubic crystal



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Surface structure description - notations

Wood's notation

$$|a_s| = m|a|$$

$$S(hkl) - i(m \times n)R\phi^\circ - N Ad$$

(p or c)

$$|b_s| = n|b|$$

$$\text{Ni}(100) - (2\sqrt{2} \times \sqrt{2})R45^\circ - O$$

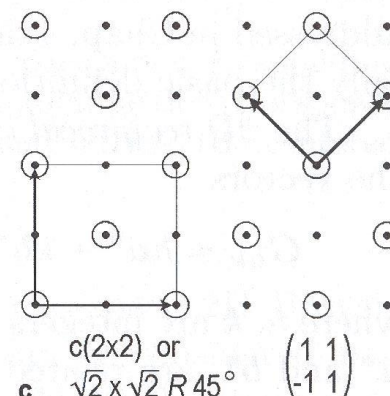
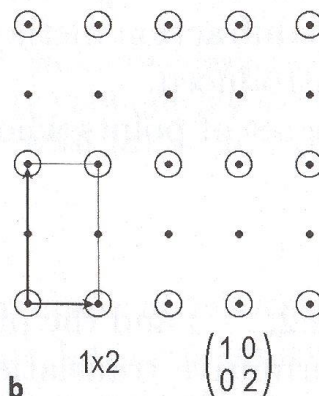
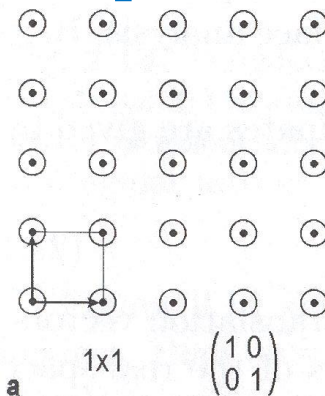
Matrix notation

$$a_s = G_{11}a + G_{12}b$$

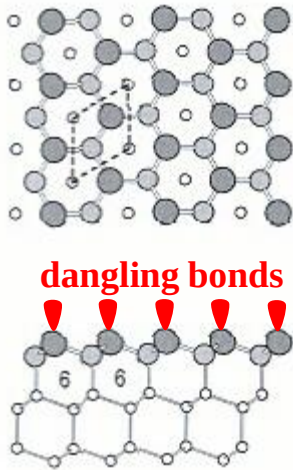
$$b_s = G_{21}a + G_{22}b$$

$$G = \begin{pmatrix} G_{11} & G_{12} \\ G_{21} & G_{22} \end{pmatrix}$$

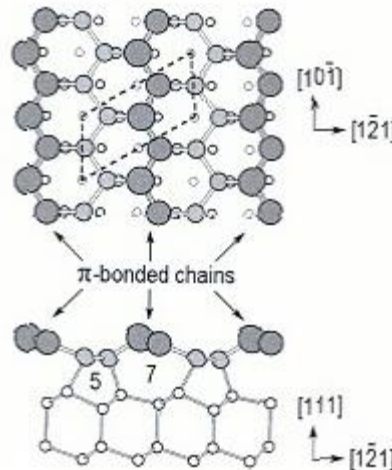
Examples:



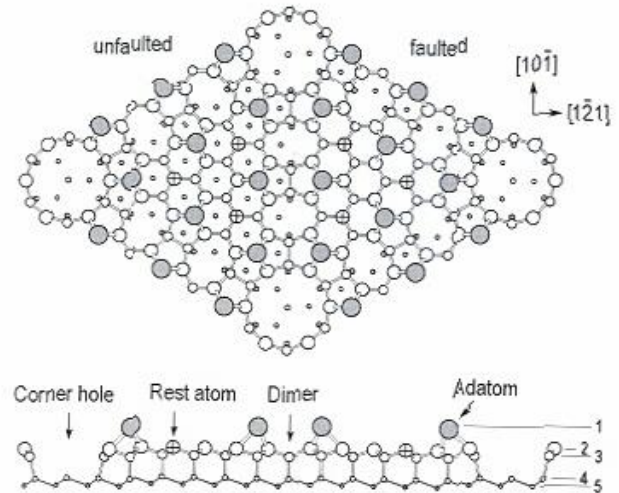
Example: Si (111) surface



Si(111)- (1x1)
ideal cut



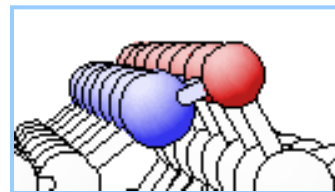
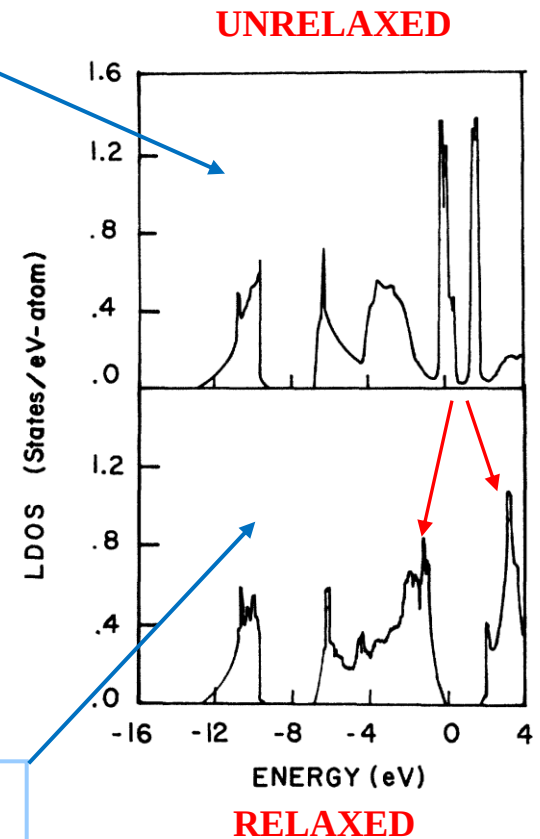
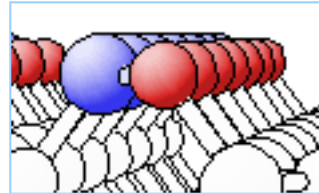
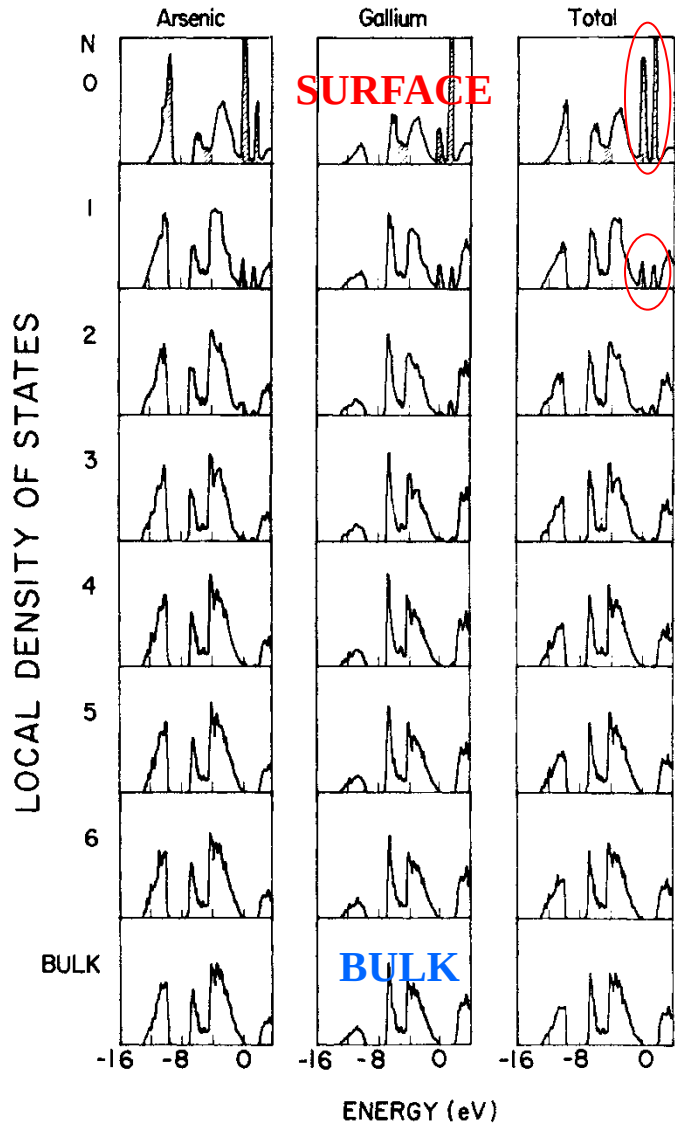
Si(111)- (2x1)
crystal cleaved
along (111)



Si(111)- (7x7)
obtained from 2x1 by
annealing at 450°C
*dimer-atom-stacking fault
(DAS) model*

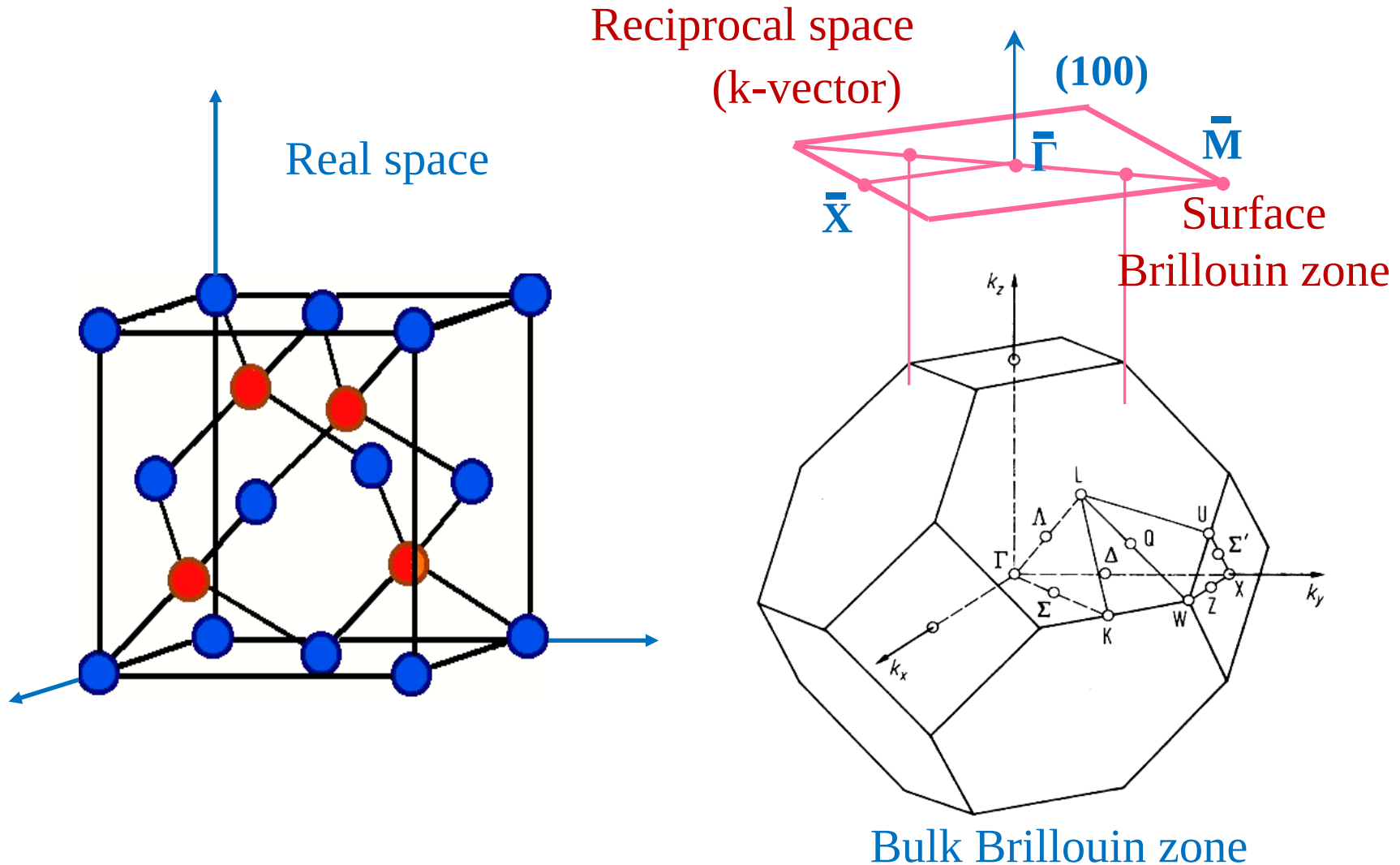
Electronic structure of the surface

unrelaxed GaAs(110)

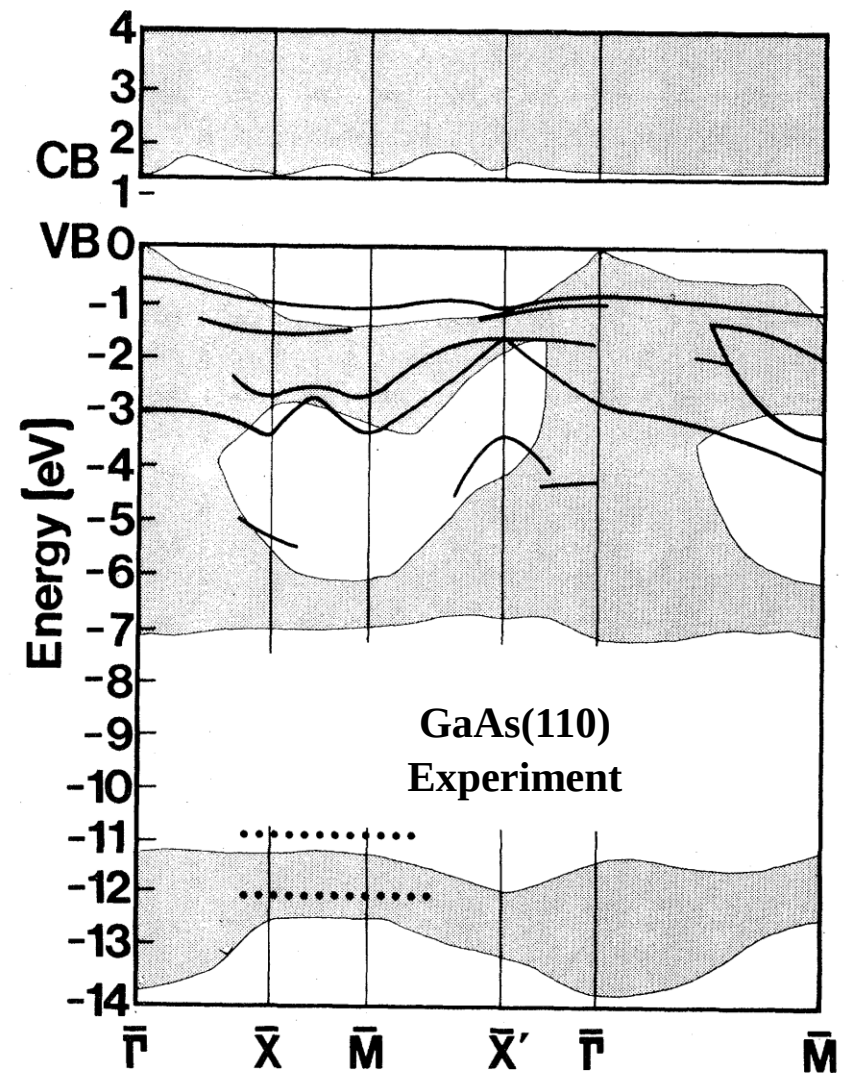
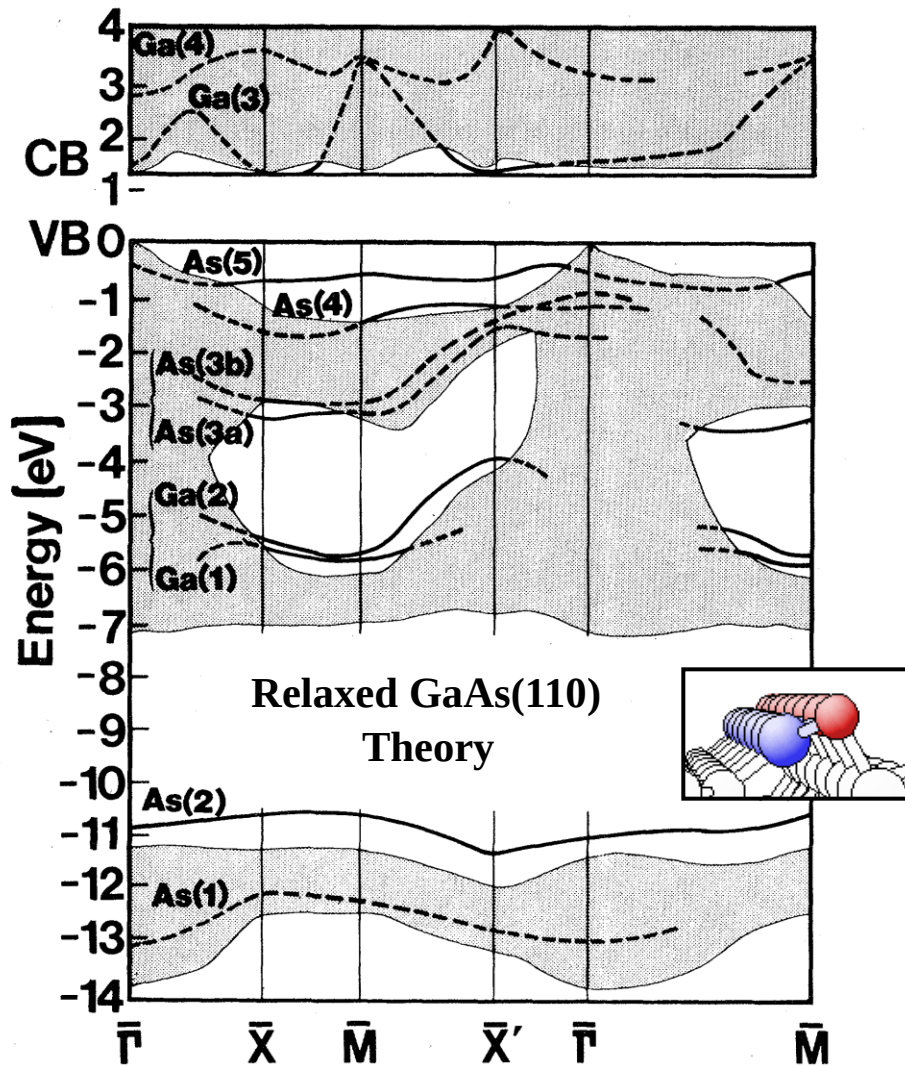


Electronic structure of the surface (cont.)

Brillouin zones



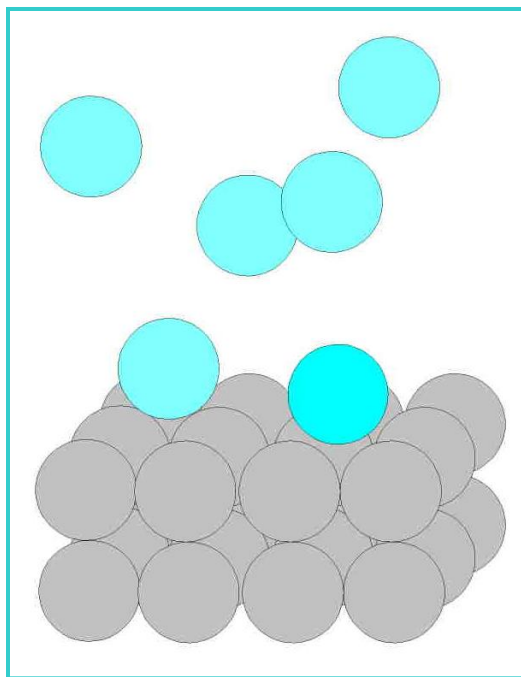
Electronic structure of the surface (cont.)



What do we want to know about surfaces?

- **Morphology**
- **Chemical composition (cleanness, presence of impurities, their surface and depth distribution...)**
- **Atomic structure**
- **Electronic structure**
- **Electronic/electric properties**
- **Optical properties**

Warning! The surface may easily be modified!



Pressure (hPa)	Mean free path	Arrival rate ($\text{cm}^{-2}\text{s}^{-1}$)	Monolayer arrival time
1000	700 Å	3×10^{23}	3 ns
10^{-3}	5 cm	4×10^{17}	2 ms
10^{-9}	50 km	4×10^{11}	1 hour

1 ML – 10^{15} cm^{-2} , sticking coefficient = 1

K.Oura et al.
Surface Science. An Introduction

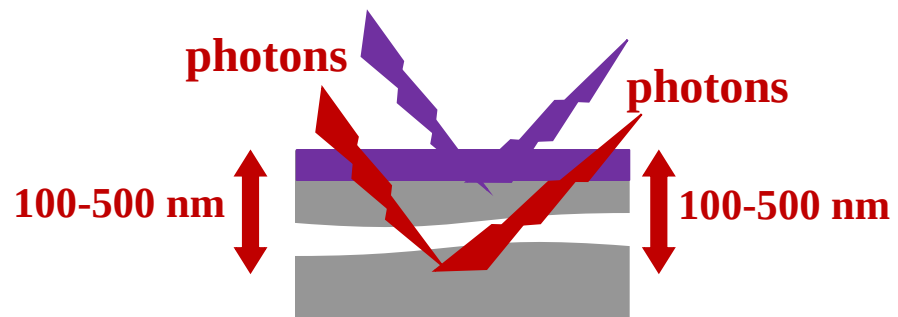
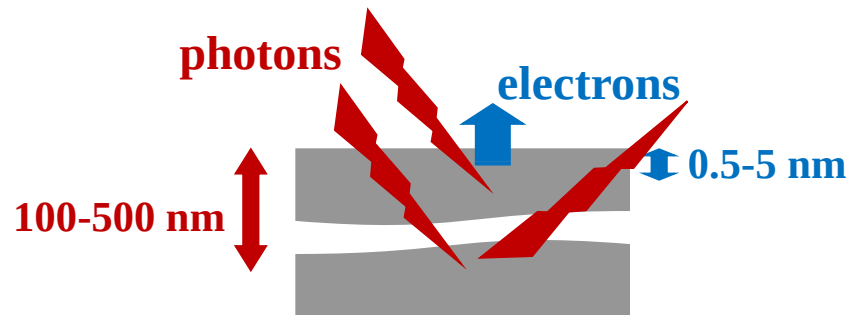
Pressure of the order of 10^{-10} hPa is necessary for studying pristine surfaces!

How to extract the signal coming from the surface?

We have to find a proper „probe”

or

a proper surface-related property



What can be a surface sensitive „probe”?

- **Electrons**

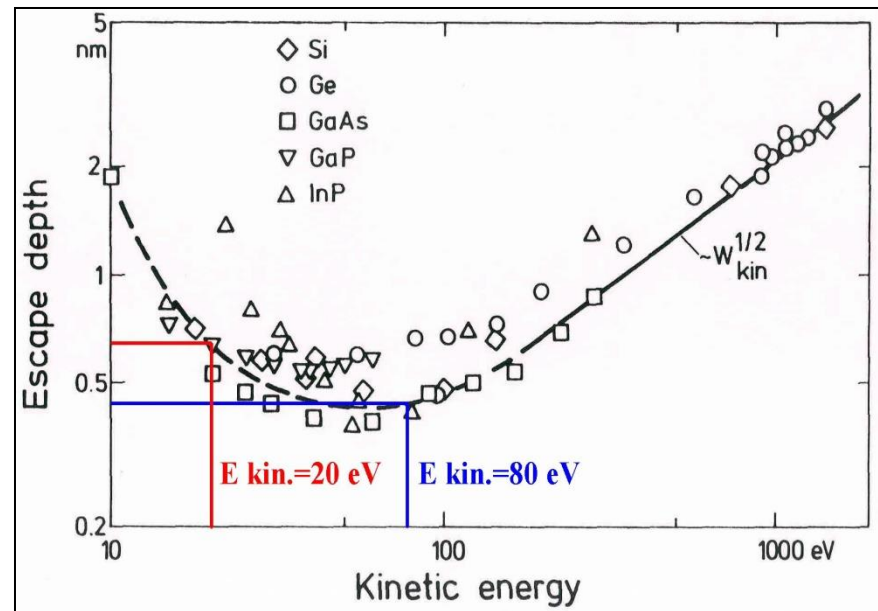
Short escape depth

Available techniques:

- **Microscopy**

- **Diffraction (LEED, RHEED)**

- **Spectroscopy (photoemission, Auger electron spectroscopy)**



W. Mönch „Semiconductor surfaces and interfaces” 1993

What can be a surface sensitive „probe” (cont)?

- **Ions**

- **Scattering (n.p. RBS)**

Increased surface sensitivity for selected crystallographic directions (channelling)

- **Surface sputtering (SIMS)**

- **Photons**

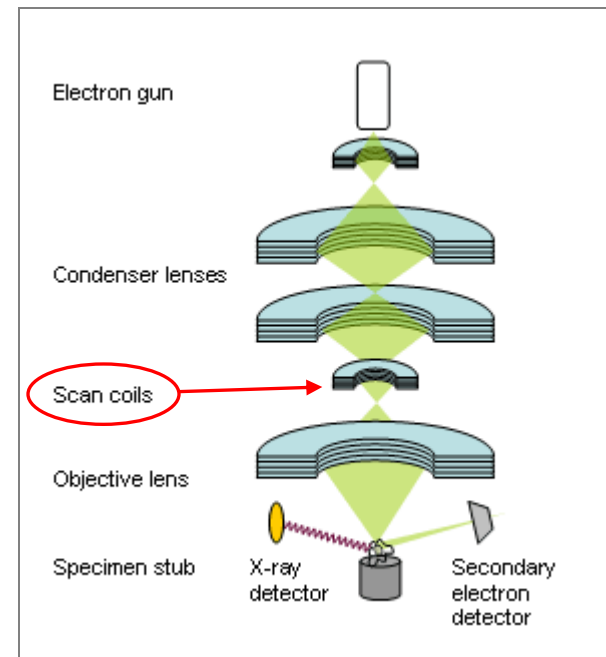
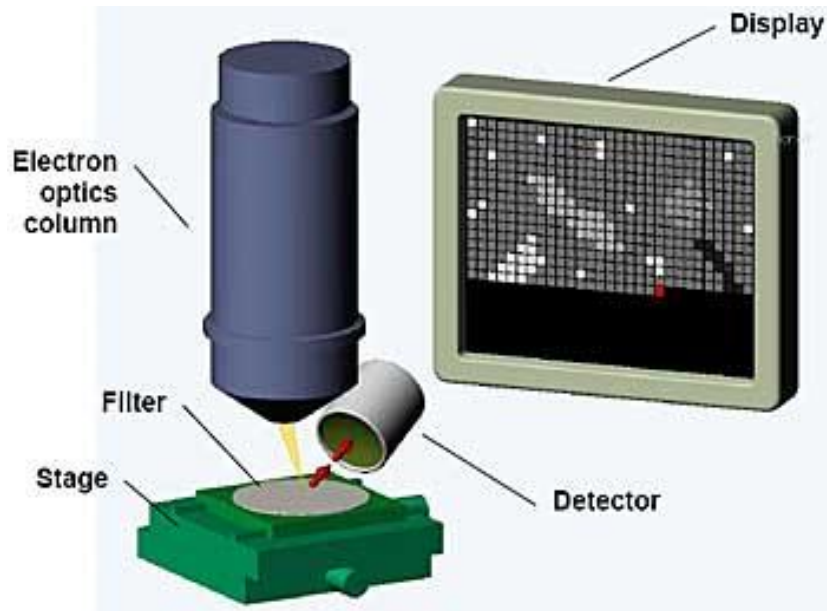
- **Surface differential spectroscopy**

- **X-ray diffraction**

Increased surface sensitivity for glancing incidence

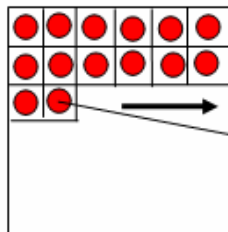
Microscopies

Scanning Electron Microscopy (SEM)

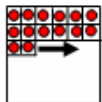


Area scanned on Specimen

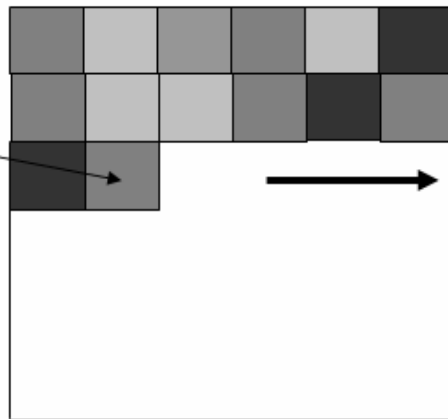
Area scanned on monitor



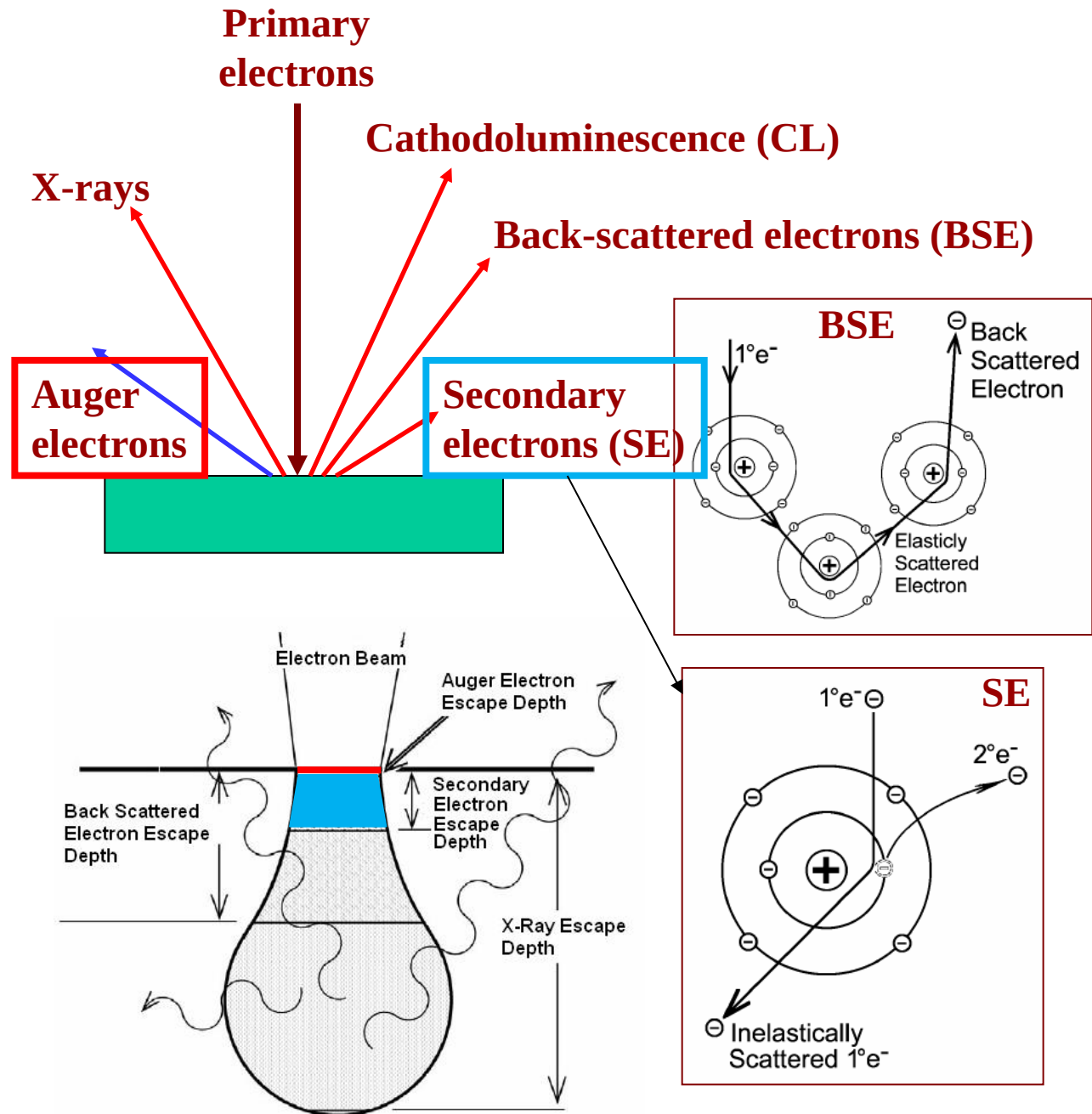
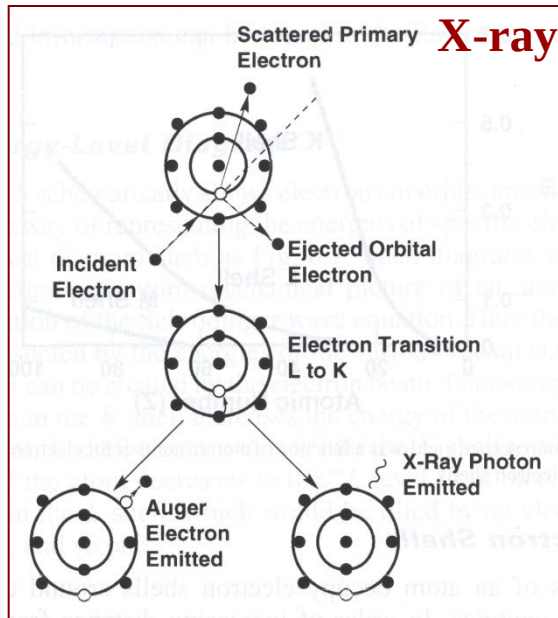
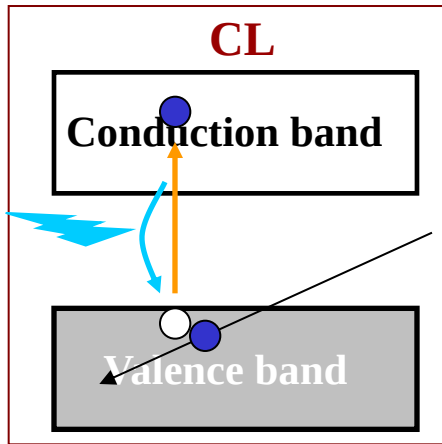
Low magnification



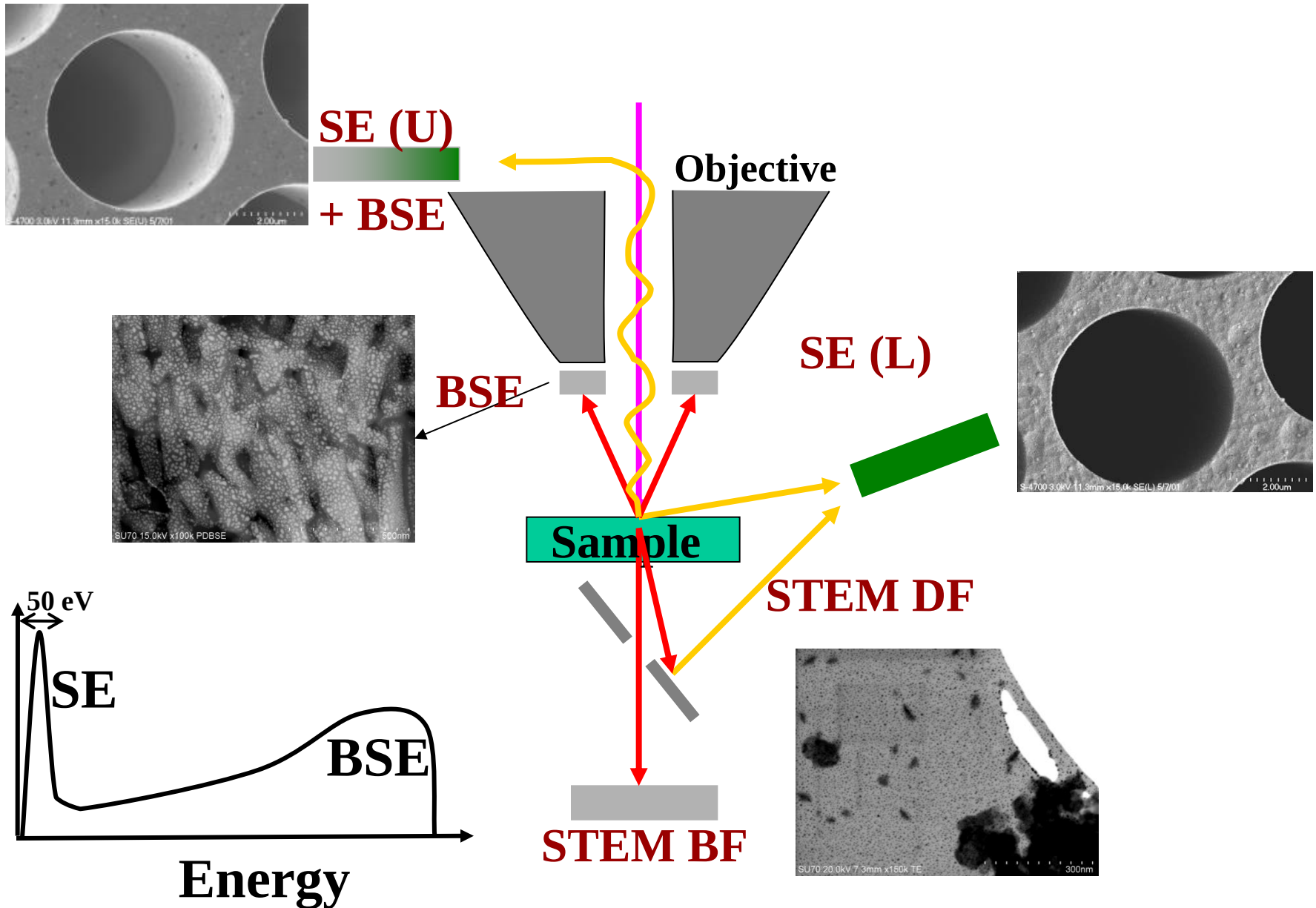
High magnification



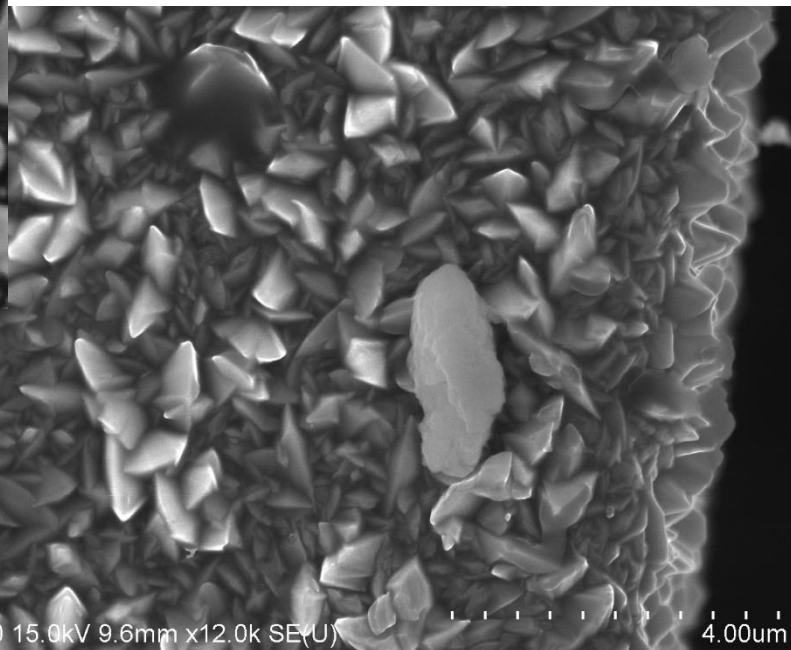
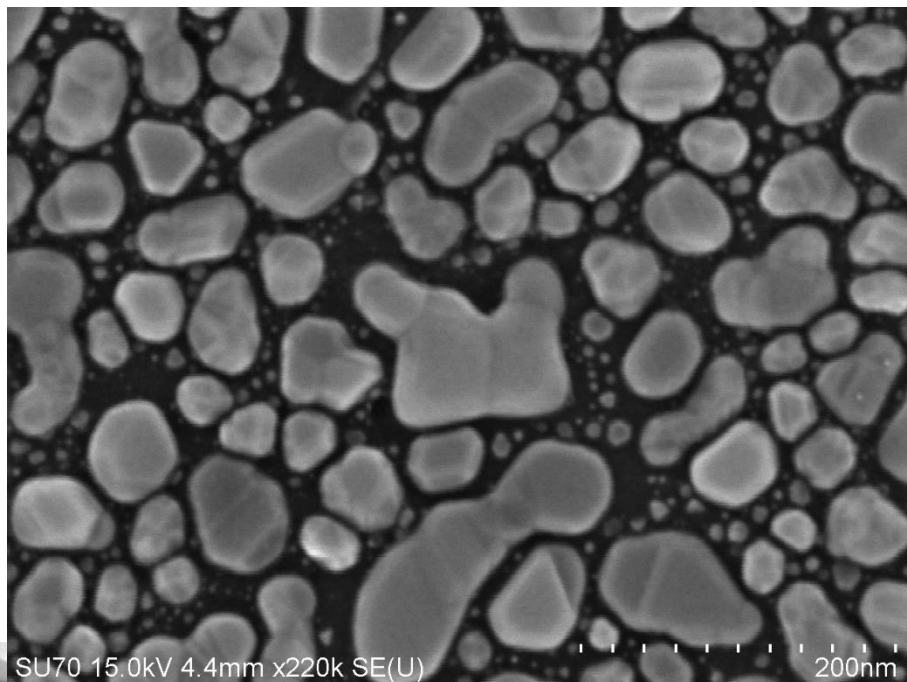
- opaque samples
- $R \approx 1 \text{ nm}$
- $U_{\text{acc}} \leq 30 \text{ kV}$



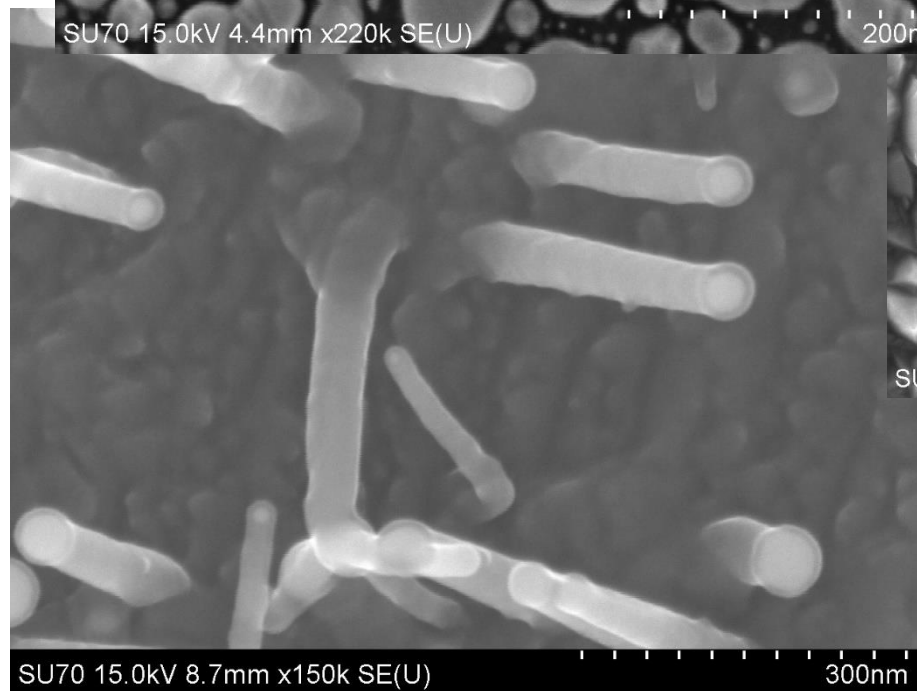
Electron detection in SEM



Au islands on C

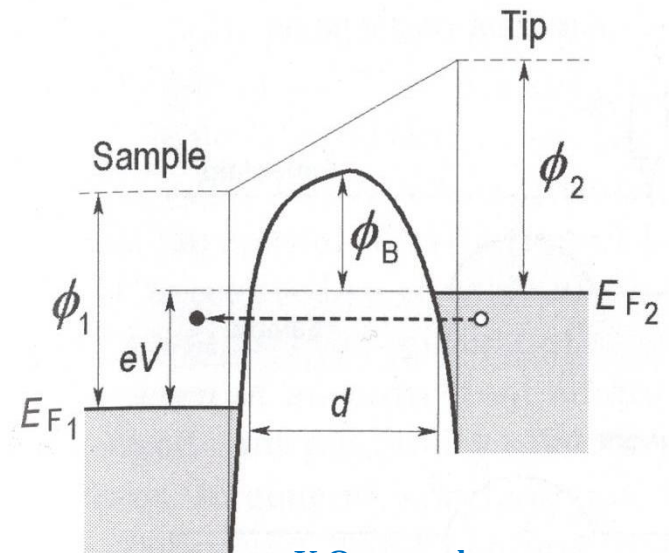
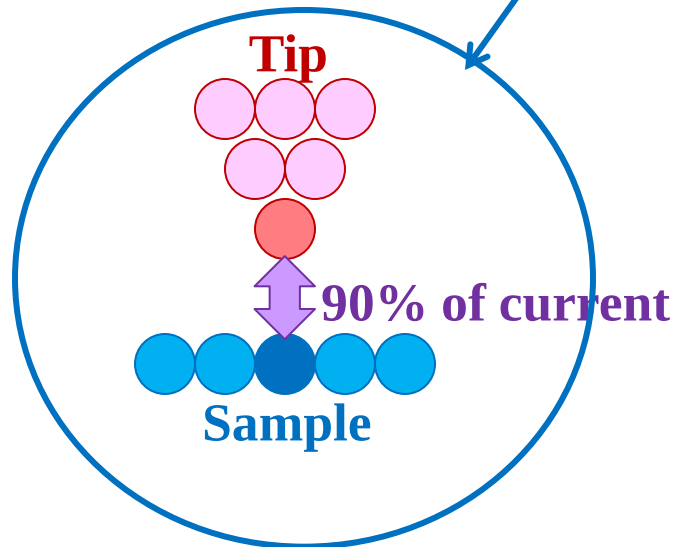
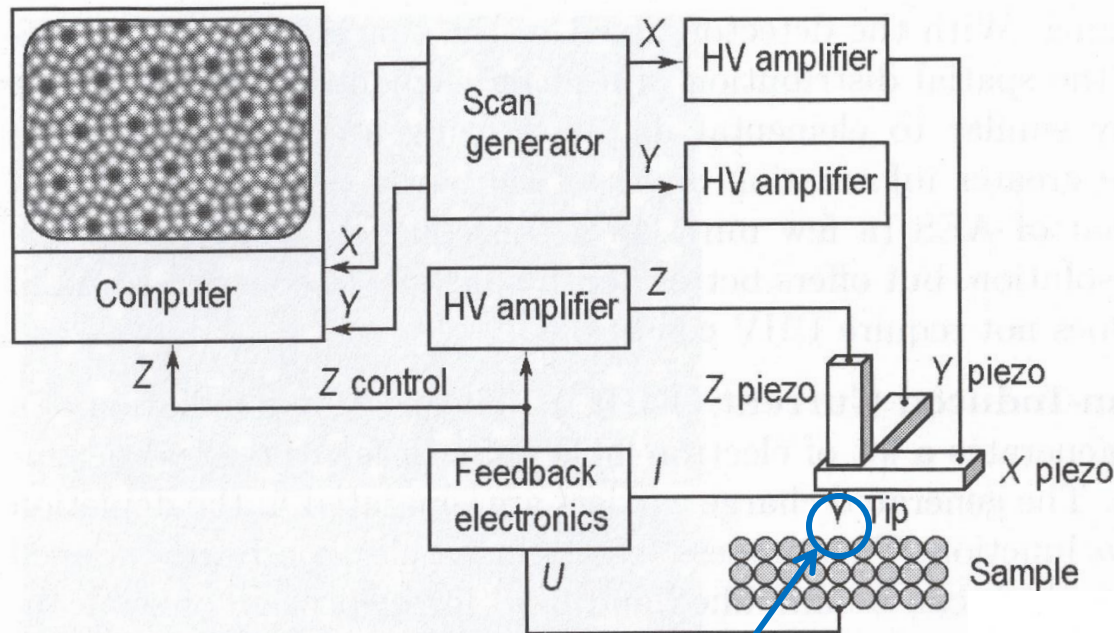


ZnO

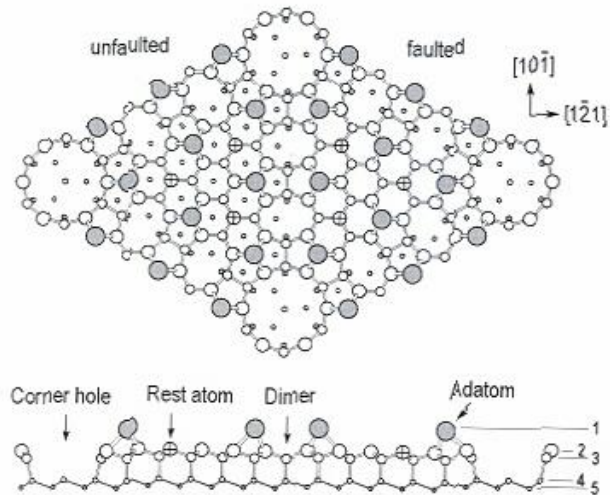


ZnTe nanowires

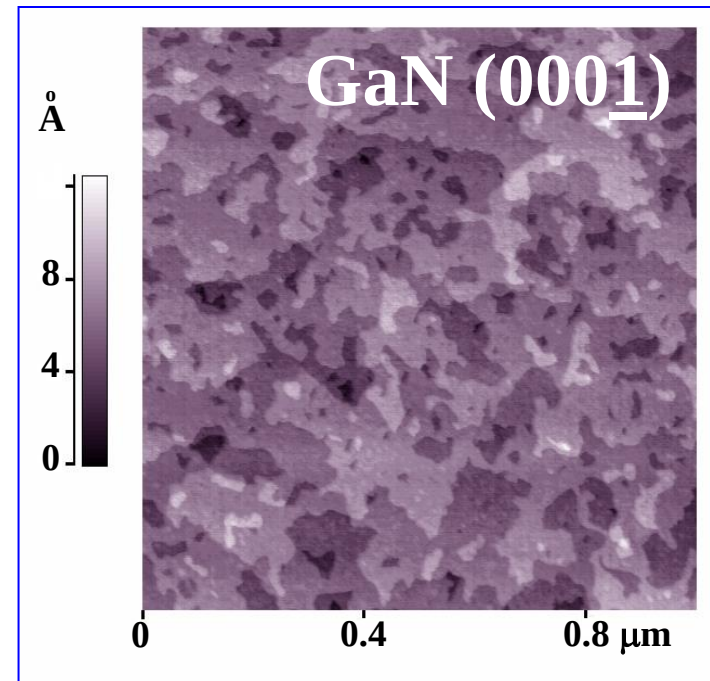
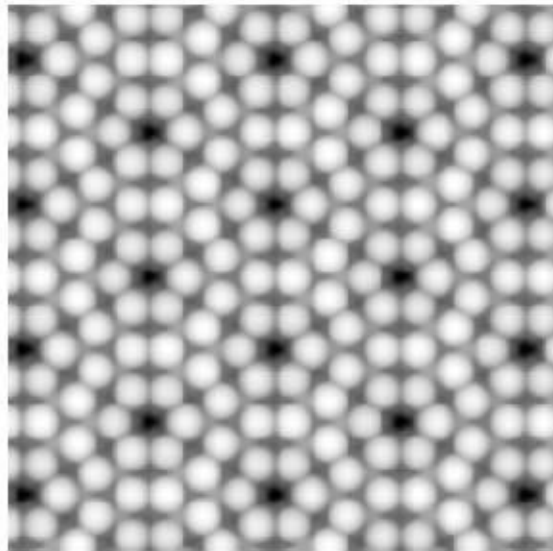
Scanning Tunnelling Microscopy (STM)



Scanning Tunnelling Microscopy (STM) (cont.)

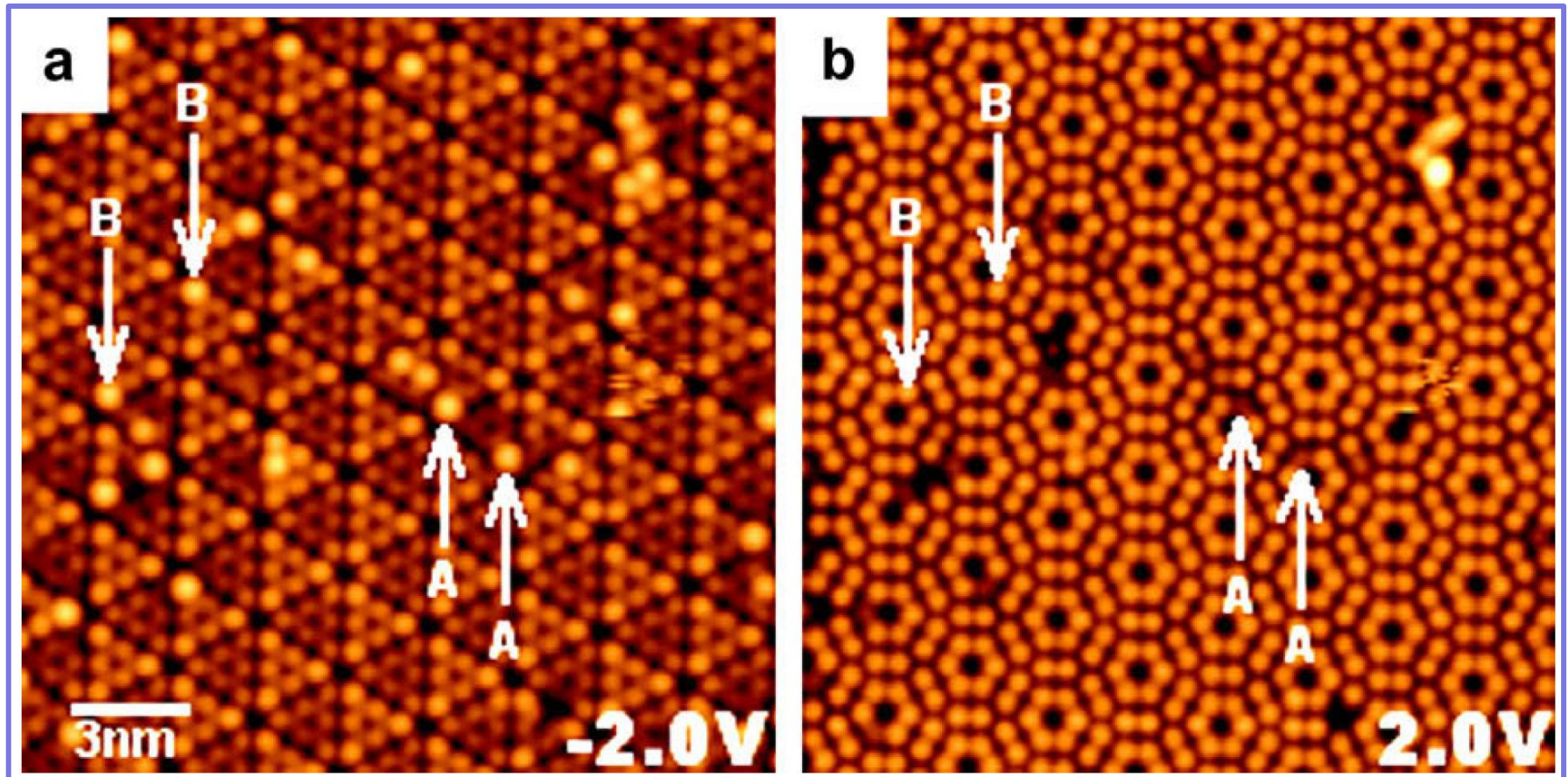


Si(111)- (7x7)



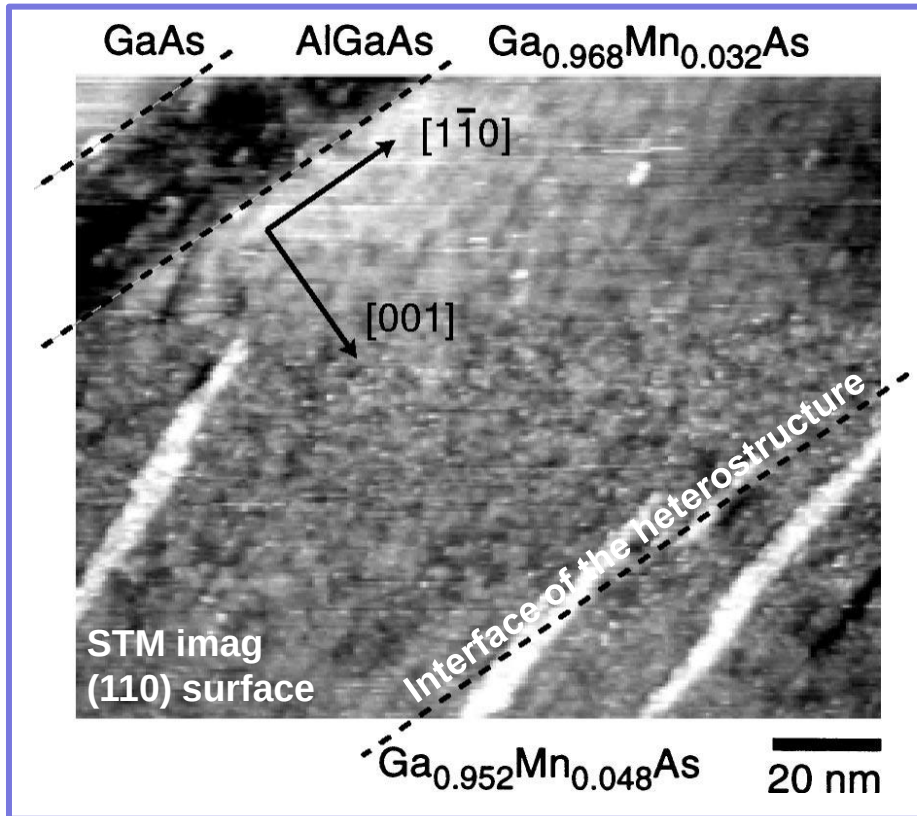
GaN(0001)- (1x1)

Scanning tunnelling microscopy (STM) (cont.)

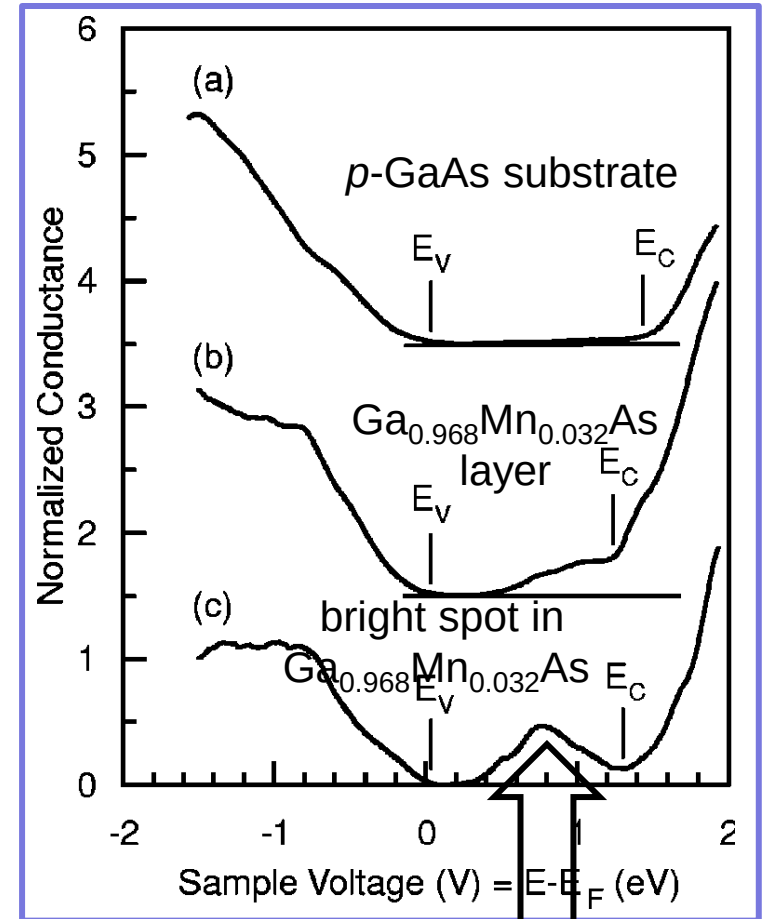


Filled and empty electronic states in STM images of Si(111)-7x7 upon deposition of 0.05 ML of Ta. (P. Shukrynau *et al.* Surface Science **603**, 469 (2009))

Scanning tunnelling spectroscopy (STS)



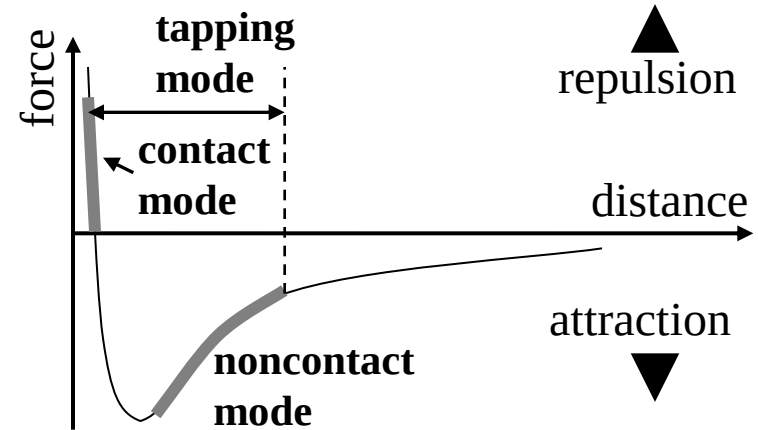
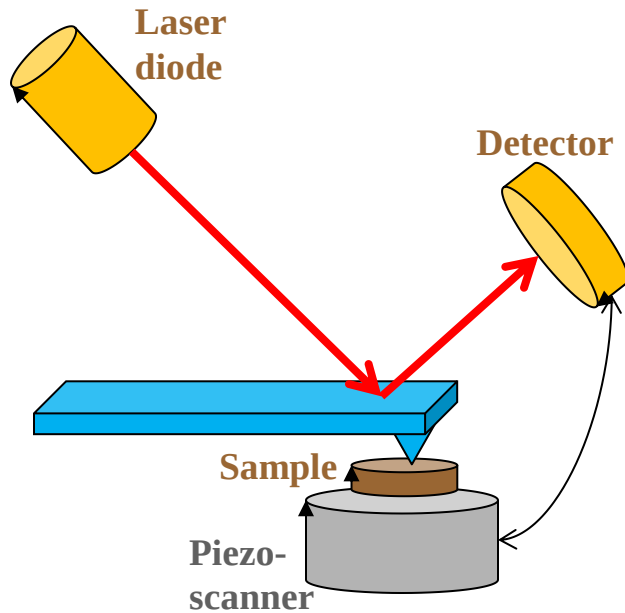
T. Tsuruoka *et al.* Appl. Phys. Lett. **81**, 2800 (2002)



Tunnelling conductance spectra (STS)

tunnelling into
the levels of As
antisites

Atomic Force Microscopy (AFM)

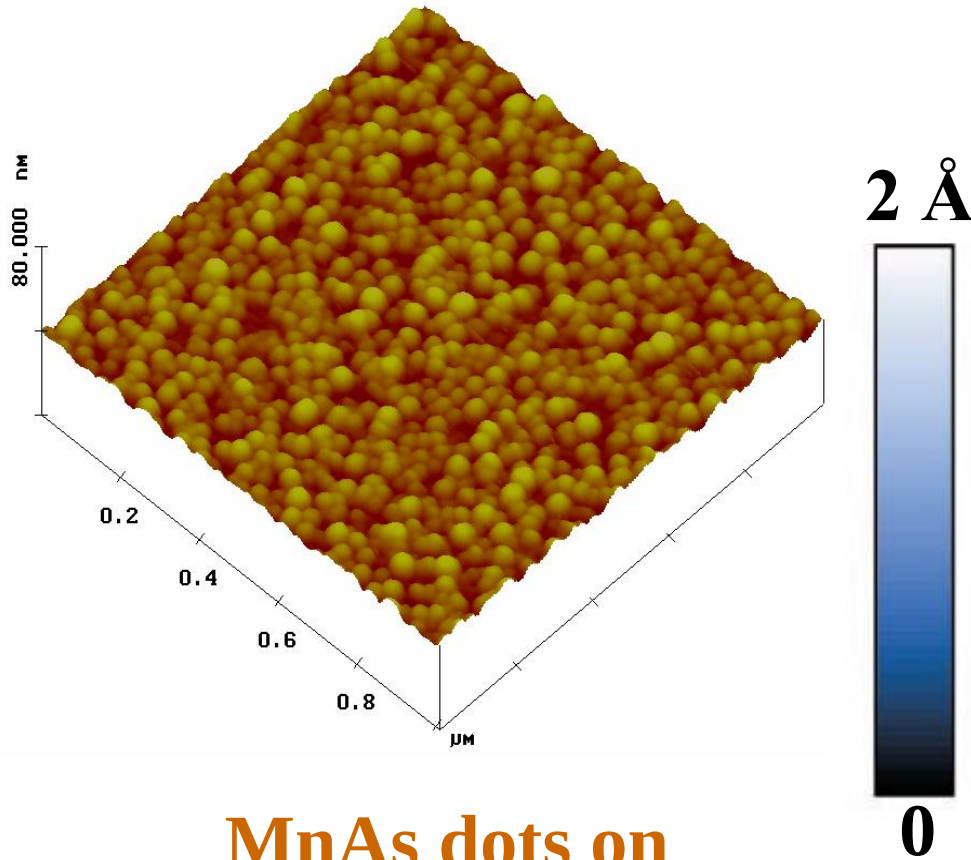


Forces:

1. Repulsive force, due to the Pauli principle ($z < 0.1$ nm).
2. Force due to the binding between atoms ($z = 0.1$ – 0.4 nm).
3. Attractive van-der-Waals force (long range, dominating for $z \gg 0.5$ nm).
4. Electrostatic forces (long range, dominating for $z \gg 0.5$ nm).
5. Attractive capillary forces (long range, larger than van-der-Waals force) additionally occur in non-UHV environments.

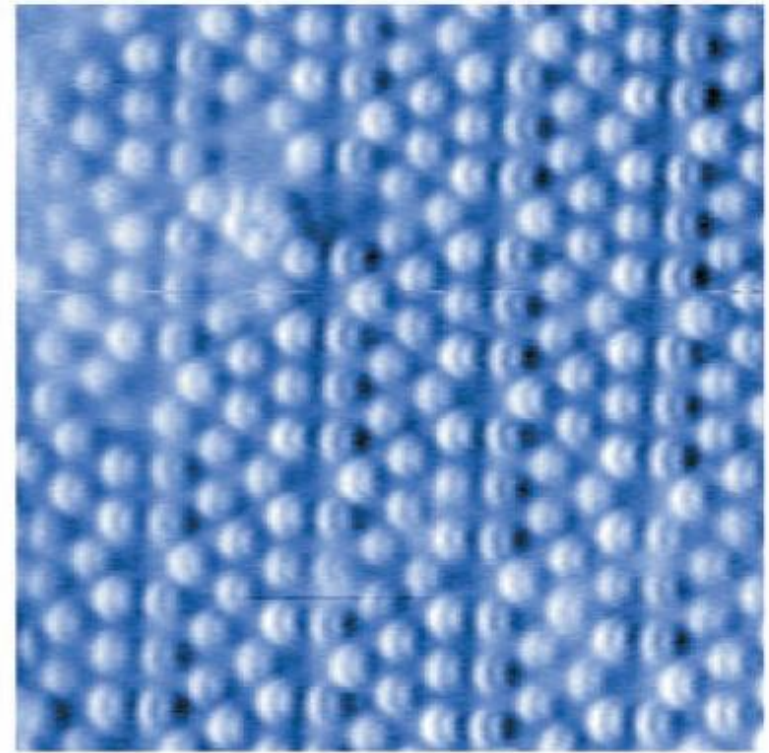


Atomic Force Microscopy (AFM) (cont.)



**MnAs dots on
GaN(0001)**

Si(111)-(7x7)



100 Å

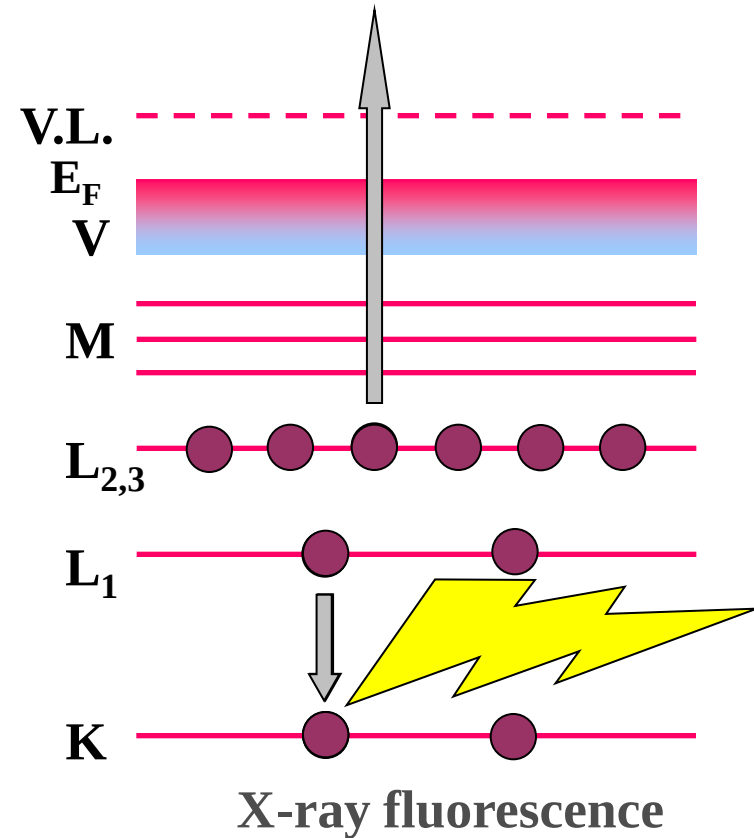
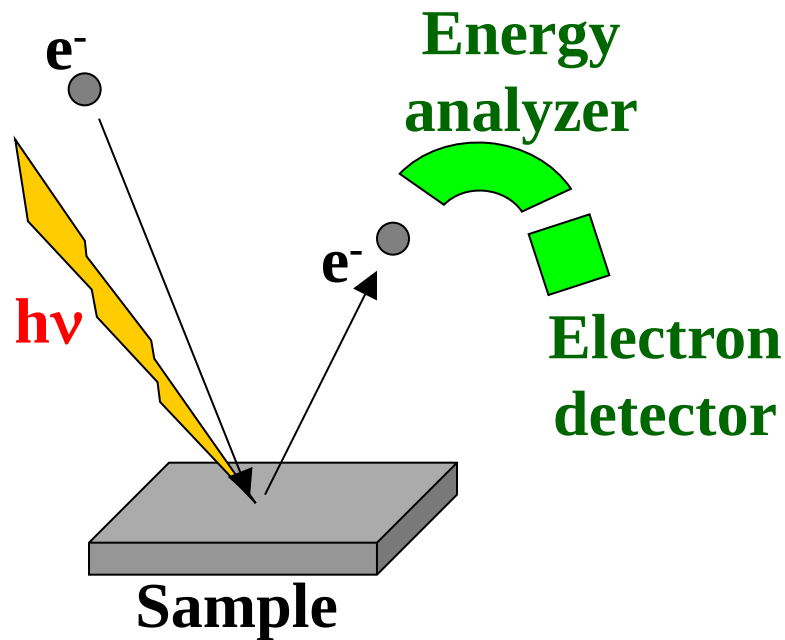
Electron spectroscopies

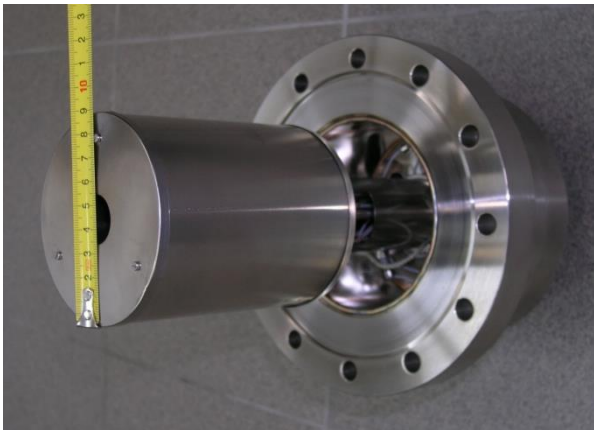
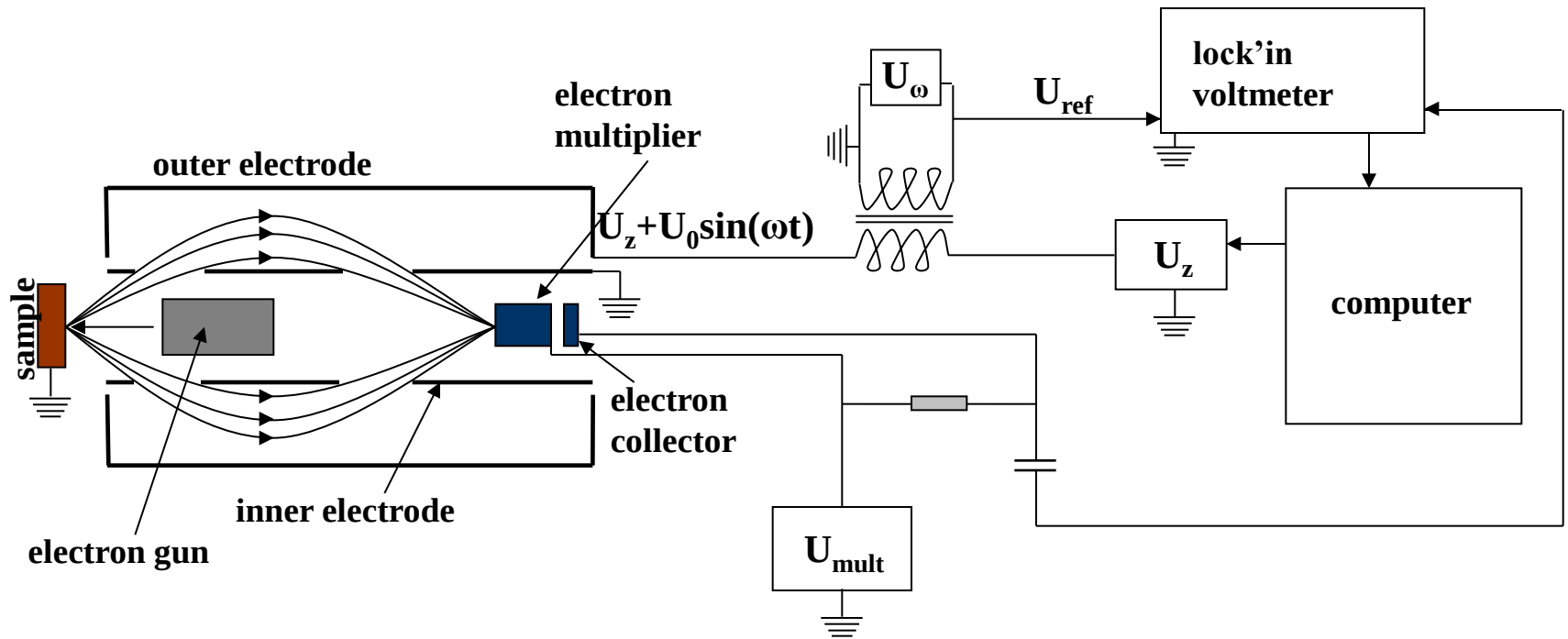
Auger electron spectroscopy

Primary electron E_0

Auger electron energy:

$$E_A = (E_K - E_{L1}) - E_{L2,3}$$





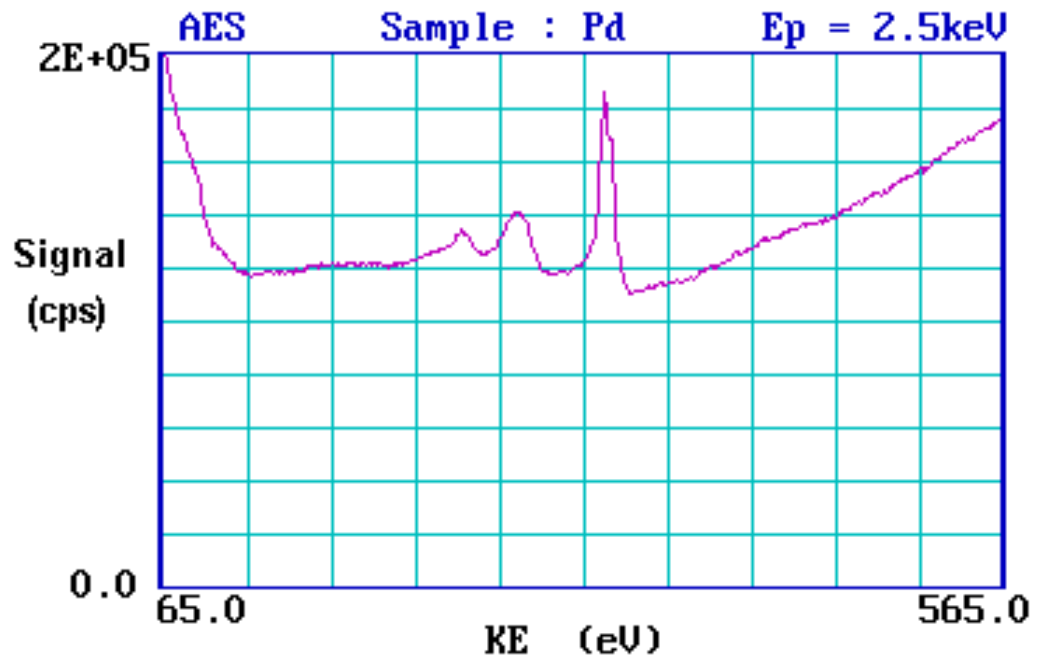
Auger electron spectrometer with a cylindrical mirror analyser

Primary electron energy: up to 3kV

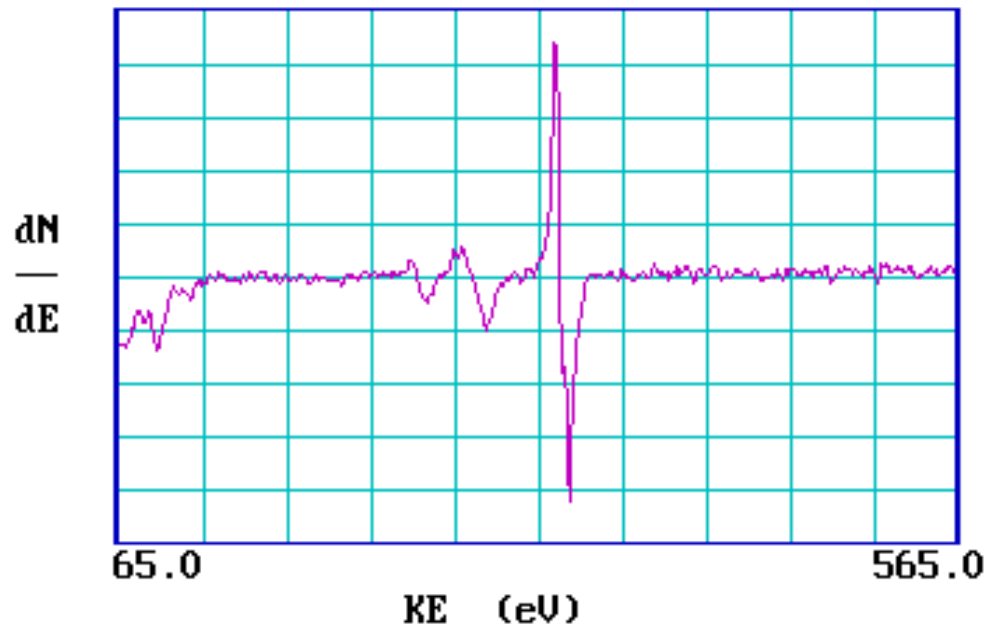
Resolving power: $E/\Delta E > 145$

Modes of AES spectra acquisition:

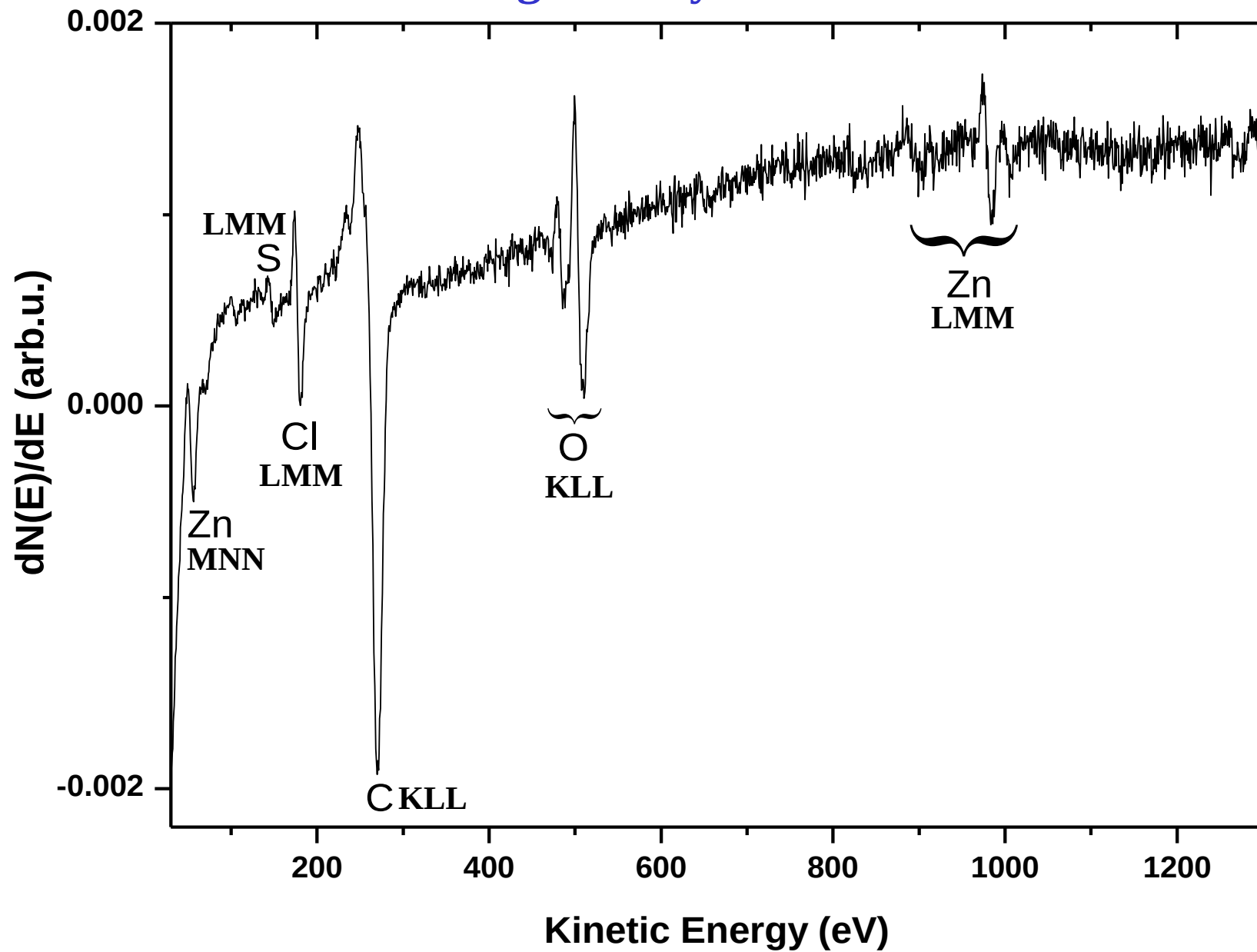
integral



differential

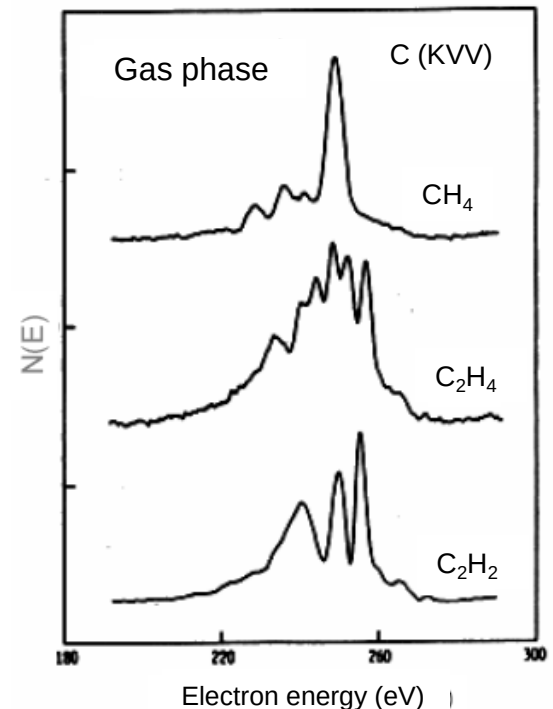


Auger electron spectrum of a ZnO layer grown by ALD

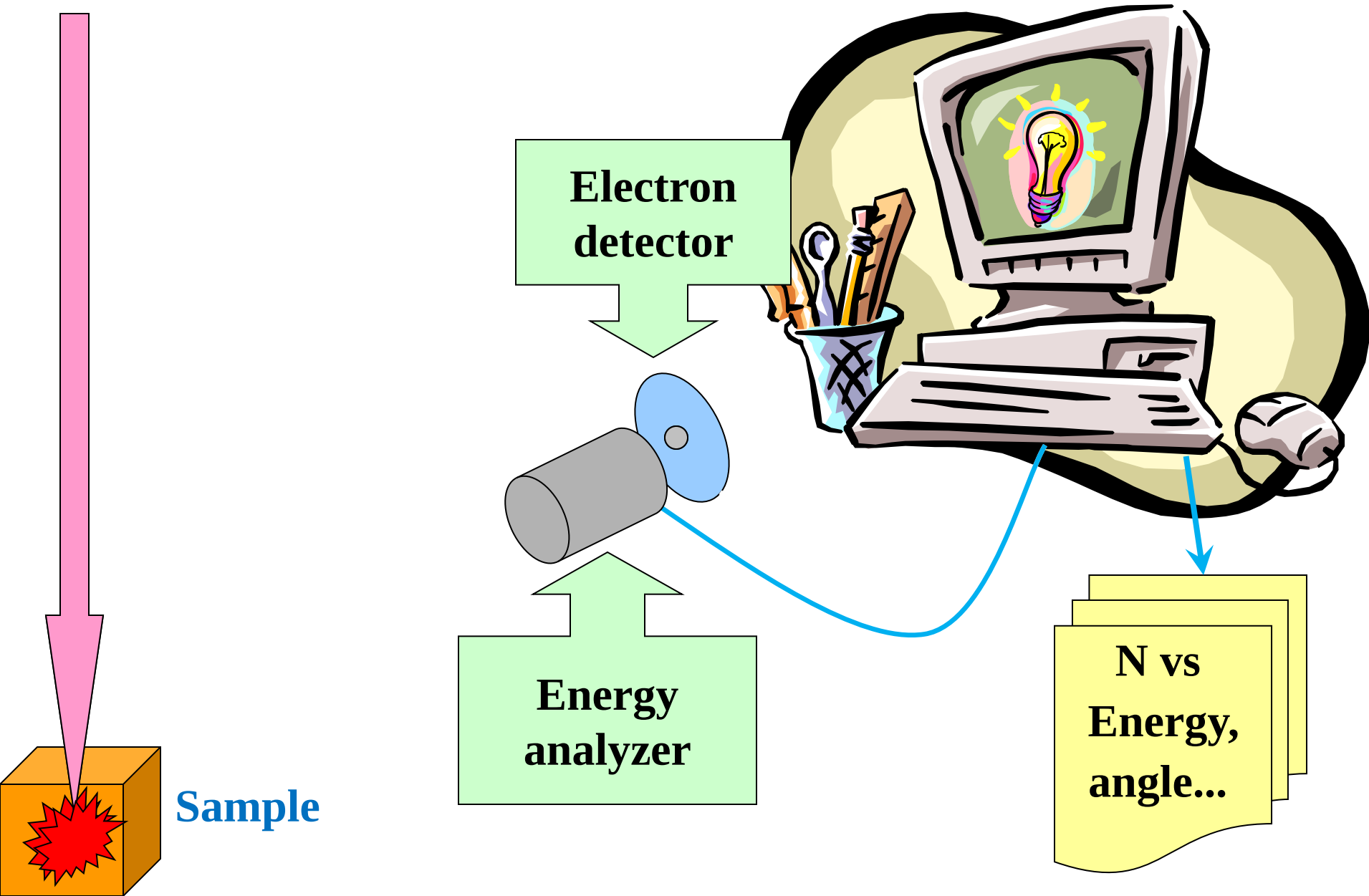


Auger electron spectroscopy:

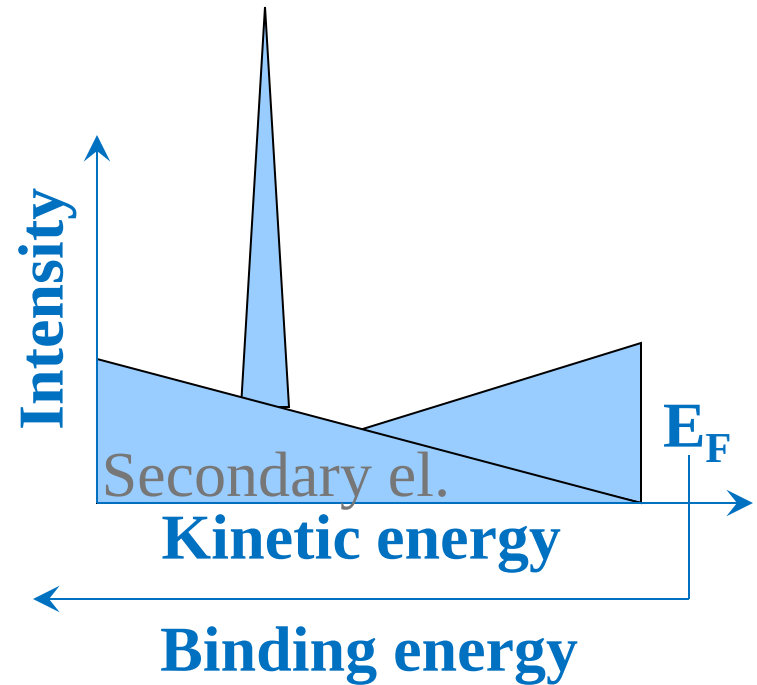
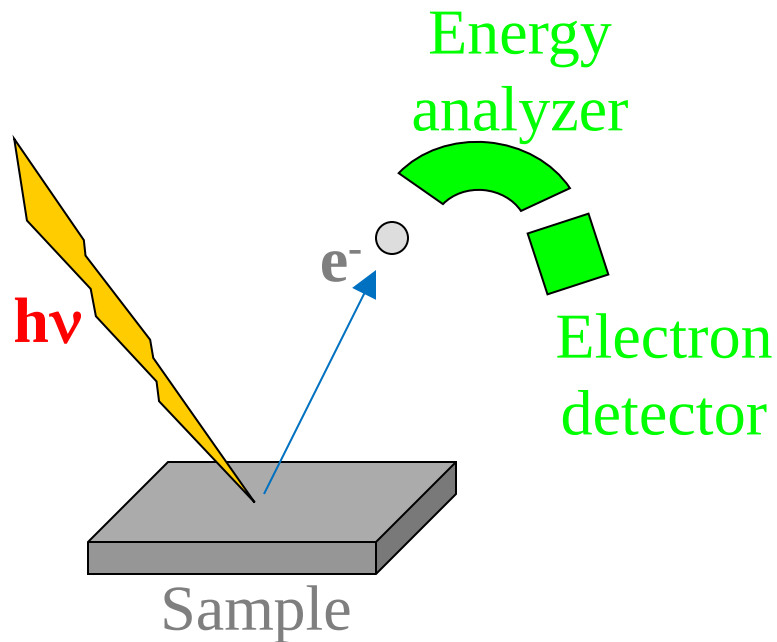
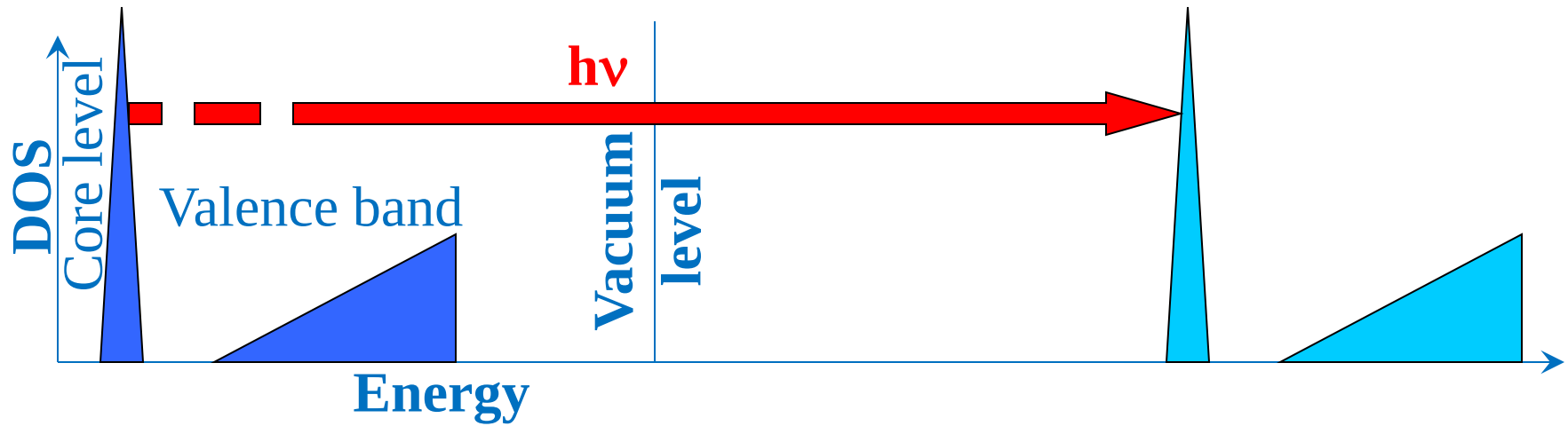
1. Analysis of the sample surface composition - detection of all elements except hydrogen and helium
2. Simple interpretation of spectra - a large database of reference spectra
3. Quantitative analysis possible - especially by comparison with standards
4. Possibility to analyze the 2D or 3D distribution
5. Sometimes spectra are sensitive to chemical bonds



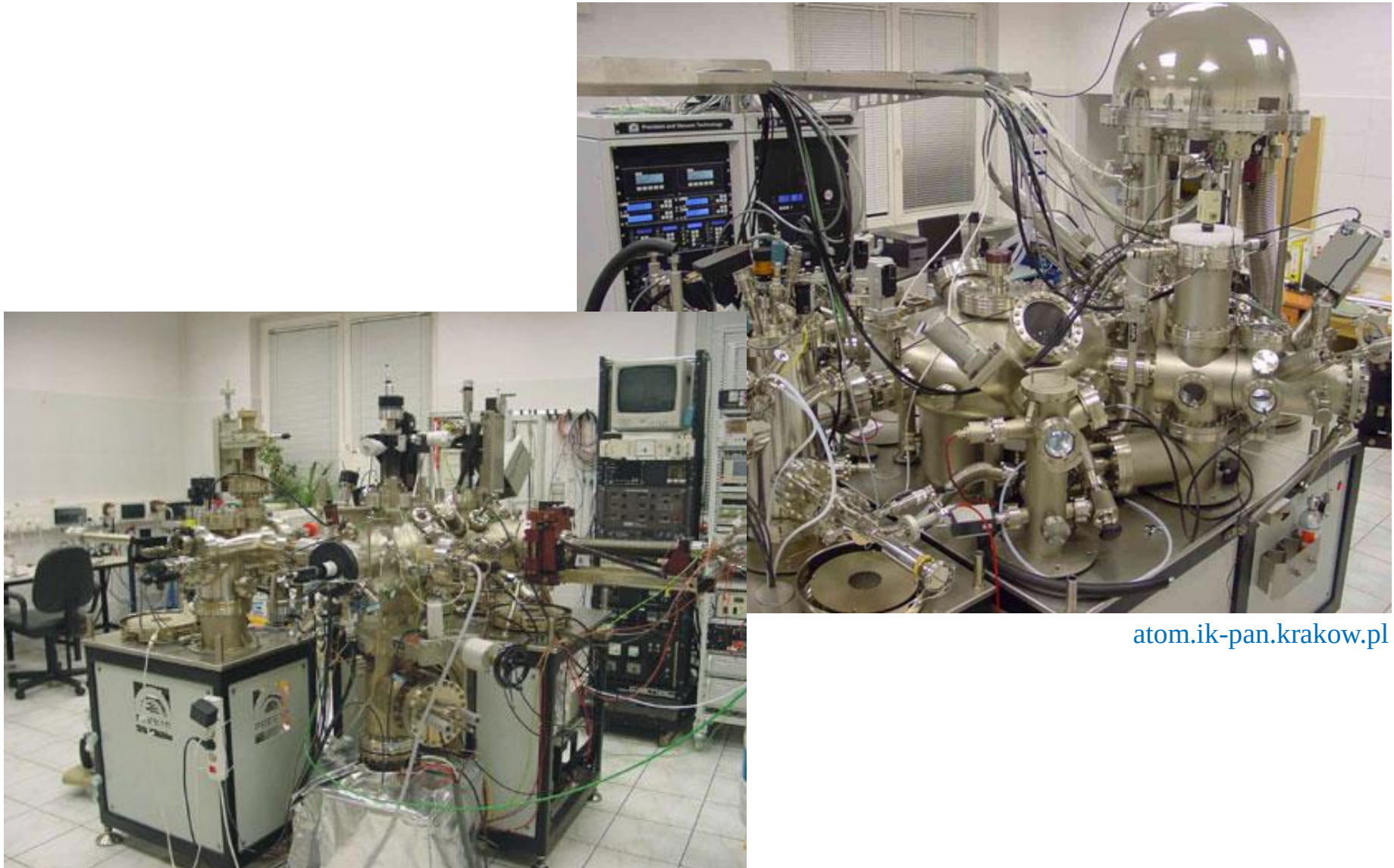
Photoemission spectroscopy



Spektroskopia fotoemisyjna



Photoemission needs Ultra High Vacuum (UHV)!



atom.ik-pan.krakow.pl

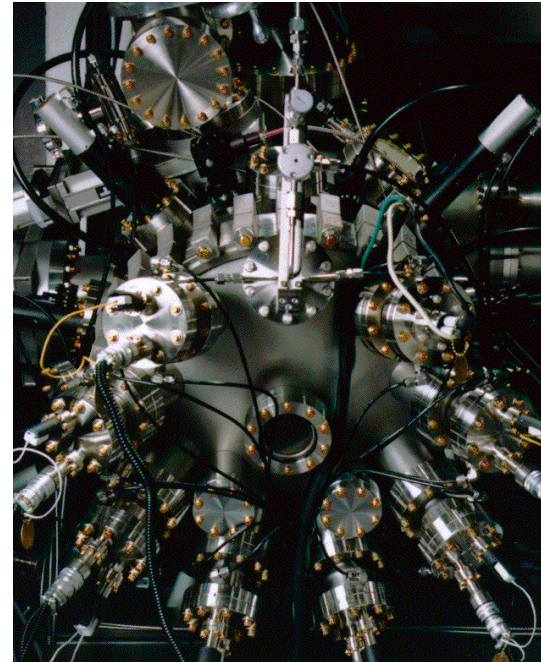
Surface preparation

www.mshel.com



□ Cleavage

□ *In situ* epitaxy

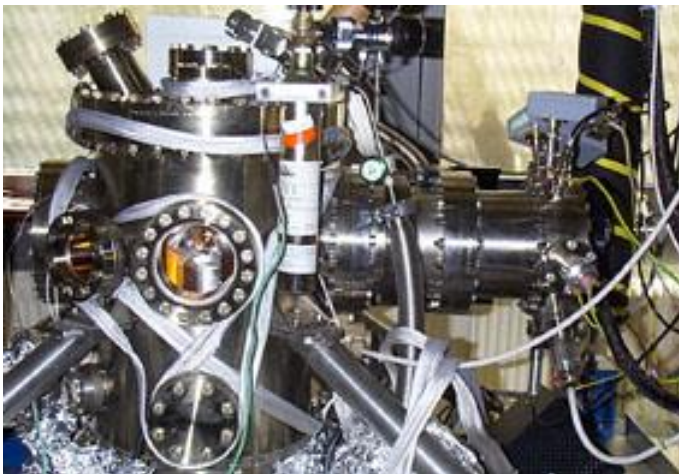


www.ems.psu.edu

□ *In situ* cleaning:

- ion etching
- annealing

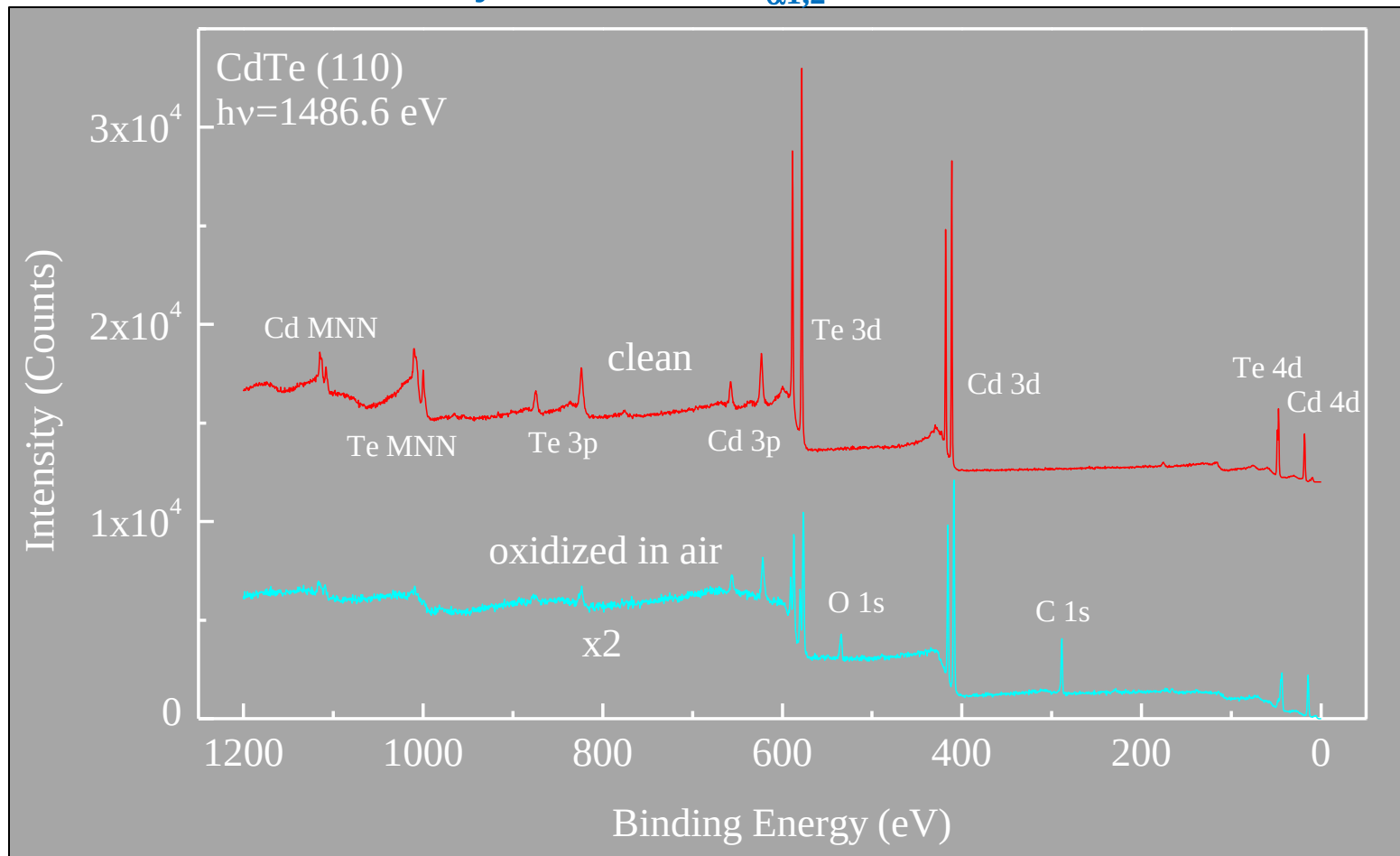
www.exphys.uni-linz.ac.at



X-ray Photoelectron Spectroscopy (XPS) or Electron Spectroscopy for Chemical Analysis (ESCA)

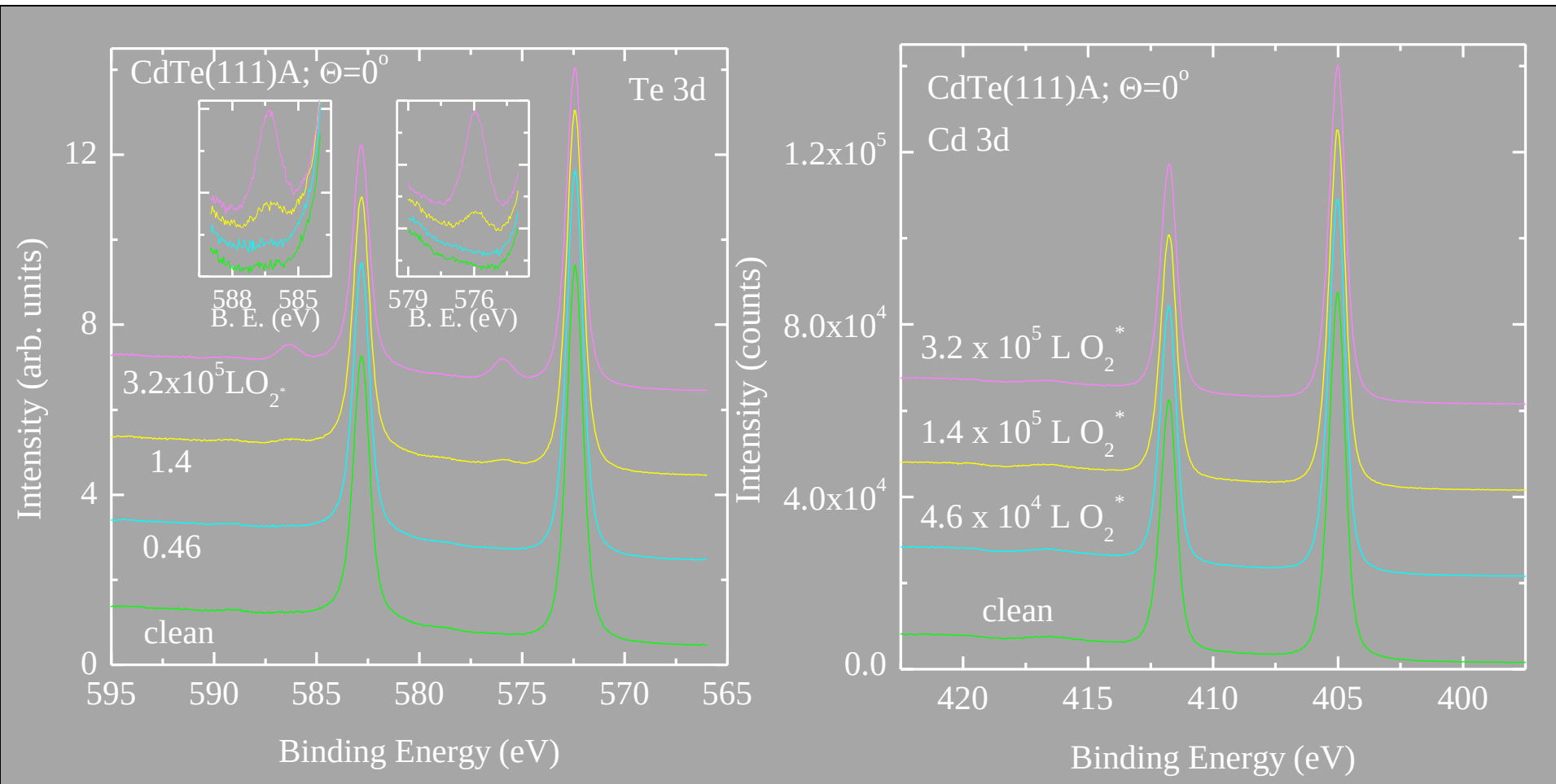
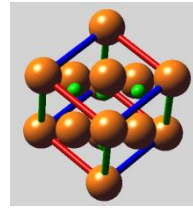
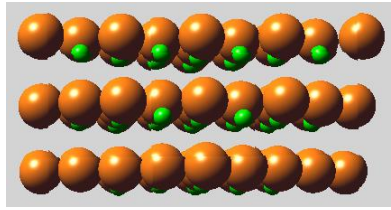
XPS: $h\nu > 1000 \text{ eV}$; $h\nu = 1000 \text{ eV} \rightarrow \mathbf{k} = 0.506 \text{ \AA}^{-1}$

X-ray source: Al $K_{\alpha 1,2}$ - 1486.6 eV



CdTe(111)A - oxidation

[111]

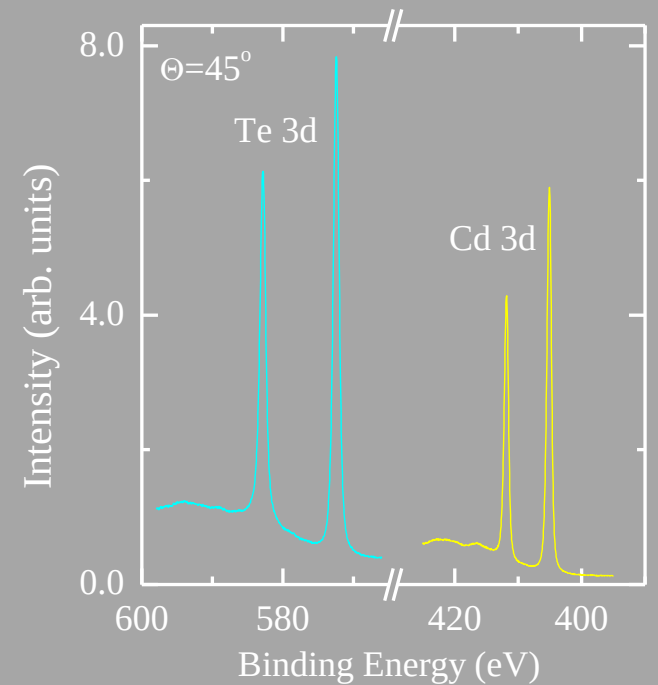
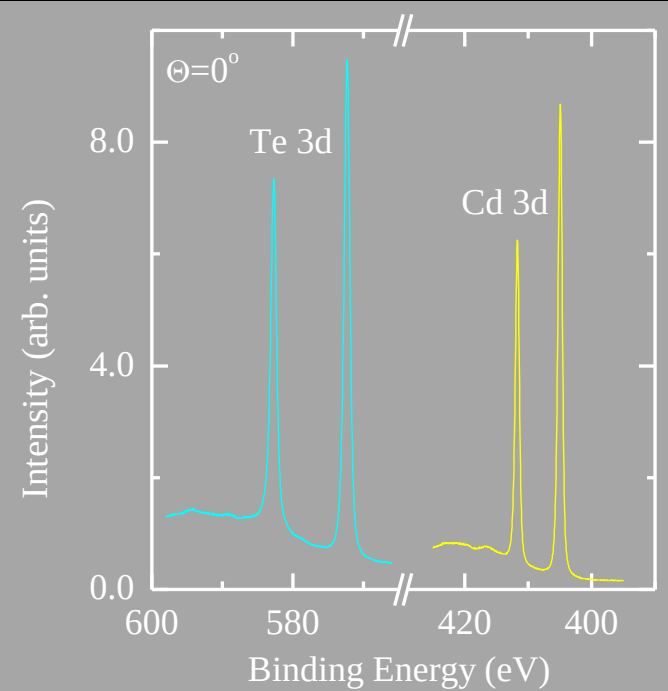
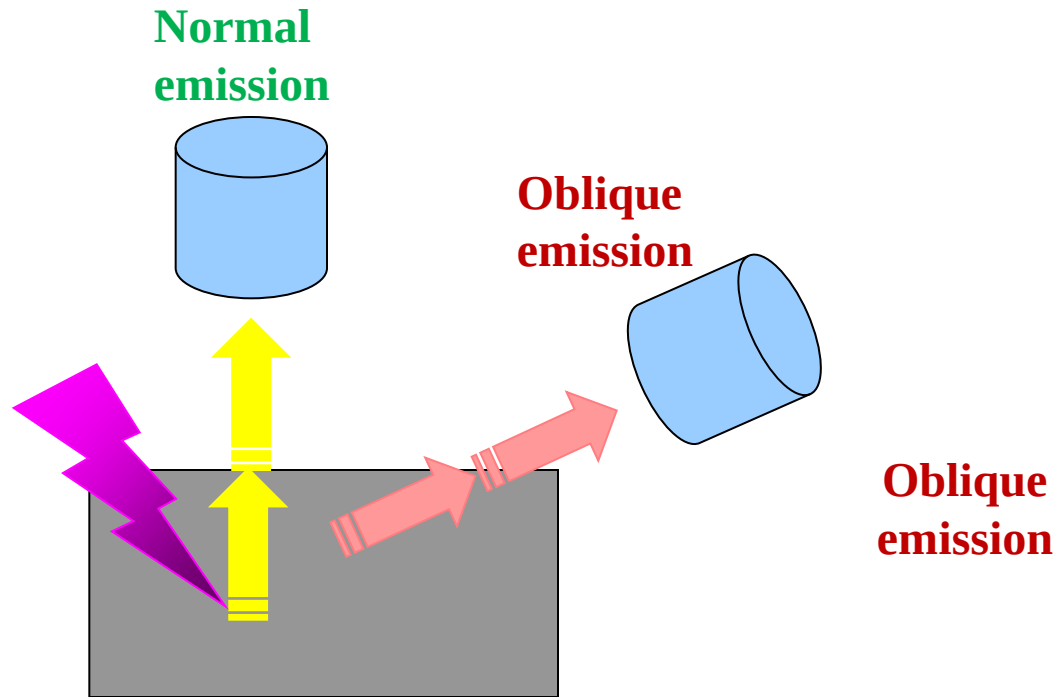


Angular XPS

- enhanced surface sensitivity

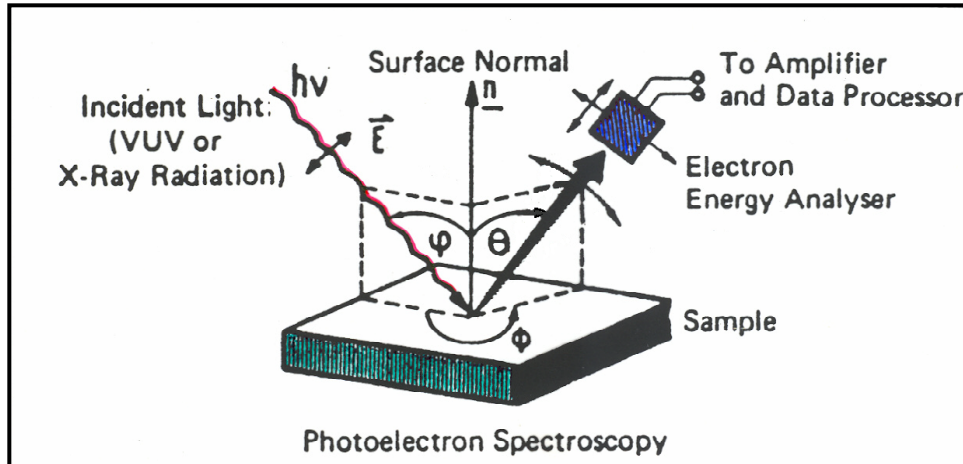
CdTe(111)A

Normal
emission

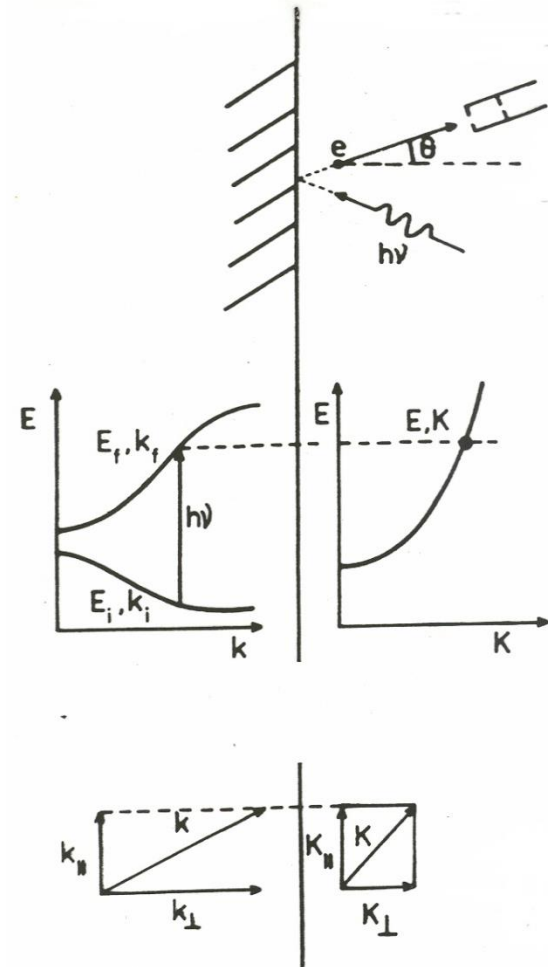


B.J. Kowalski, B.A. Orlowski, J. Ghijsen,
Appl. Surf. Sci. **166**, 237 (2000)

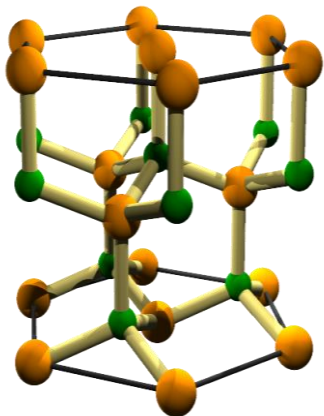
Angle-resolved photoelectron spectroscopy



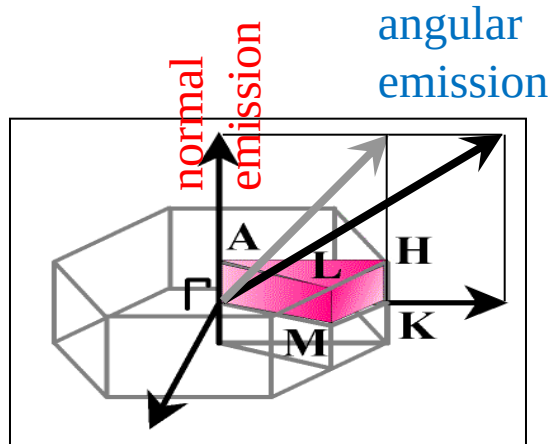
Crystal Vacuum



Example:

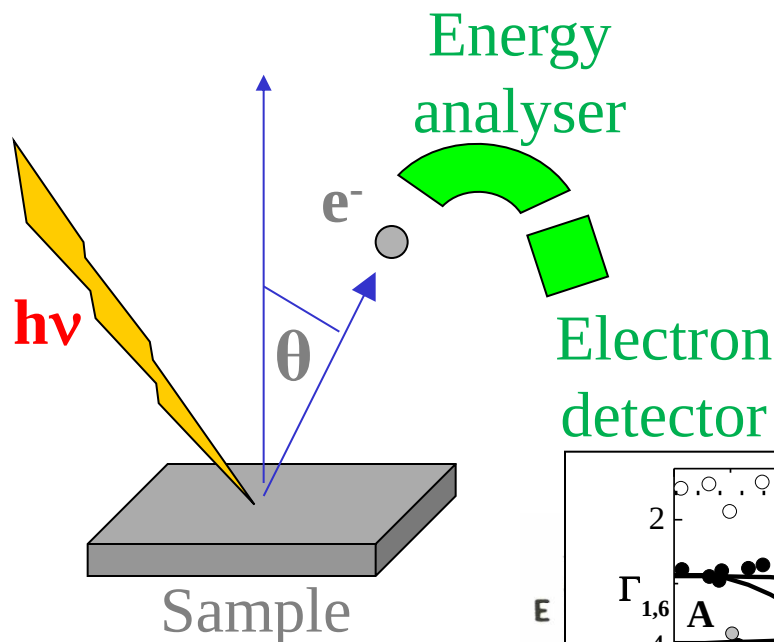
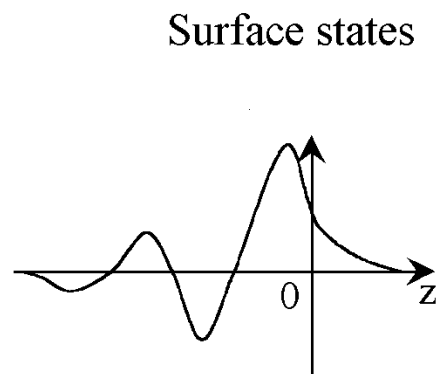
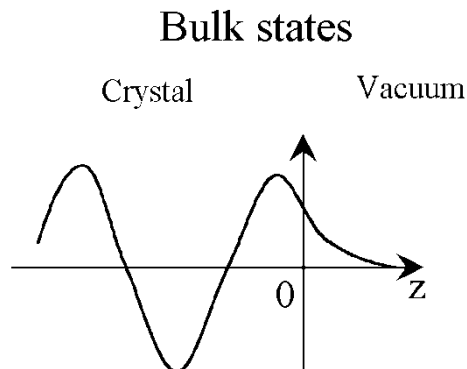


Wurtzite structure



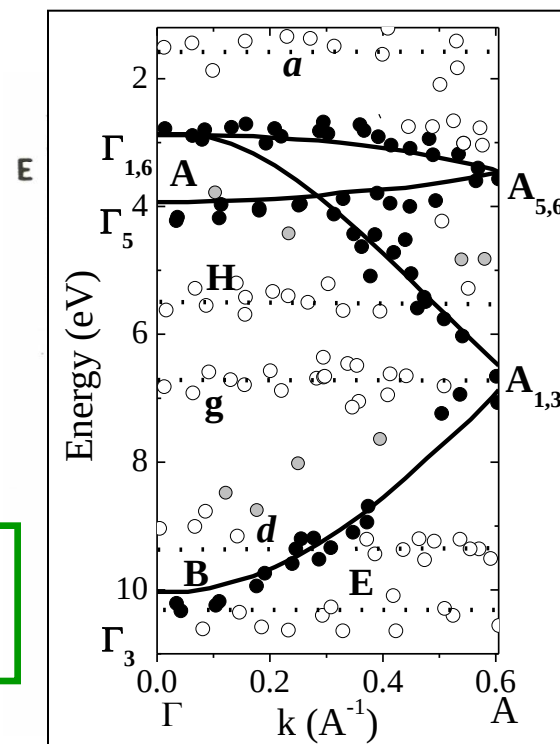
Brillouin zone

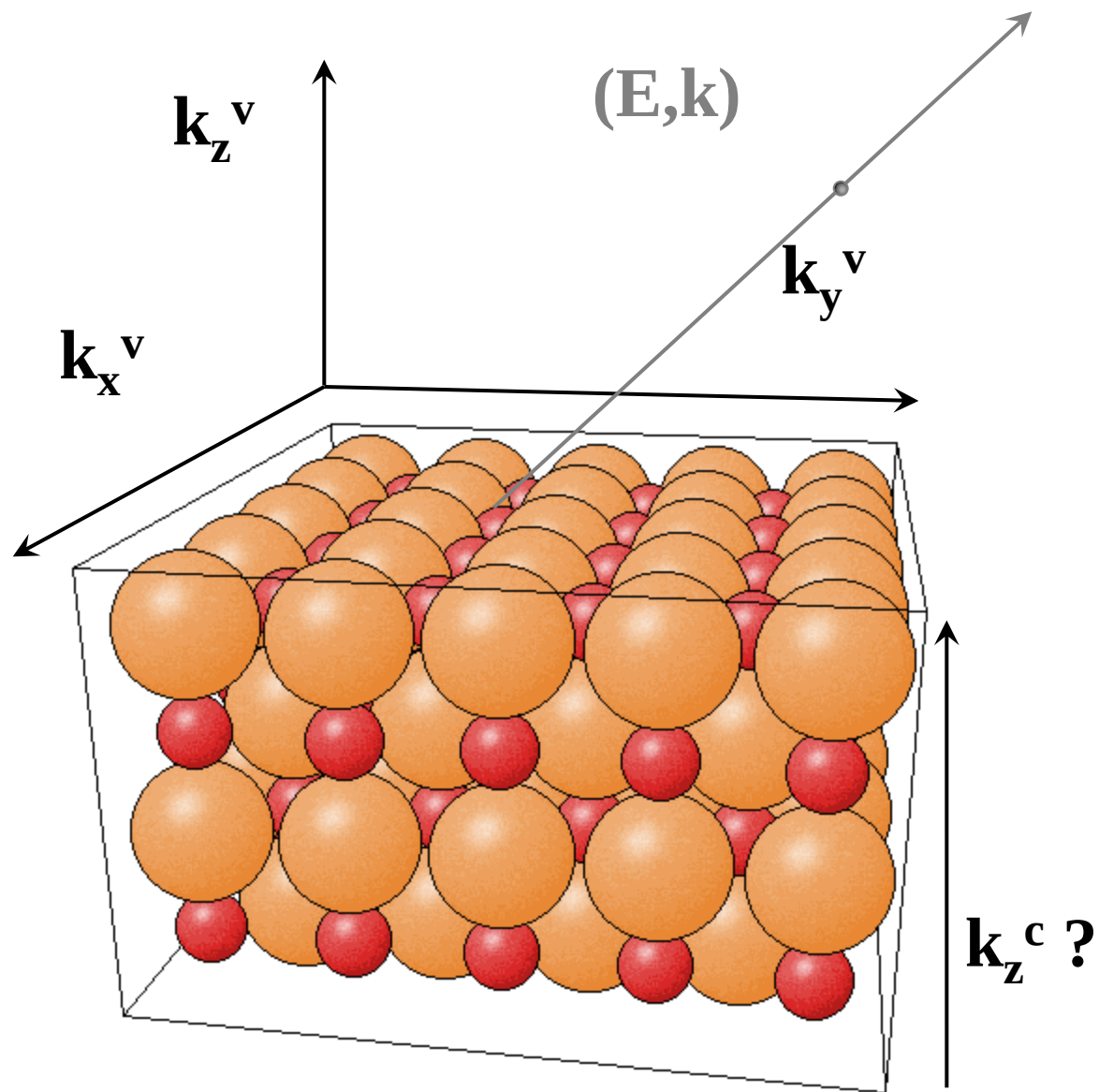
Angle-resolved photoelectron spectroscopy of surface and bulk states



$$k_{i||} = \sqrt{\frac{2m}{\hbar^2} E_{kin} \cdot \sin(\vartheta)}$$

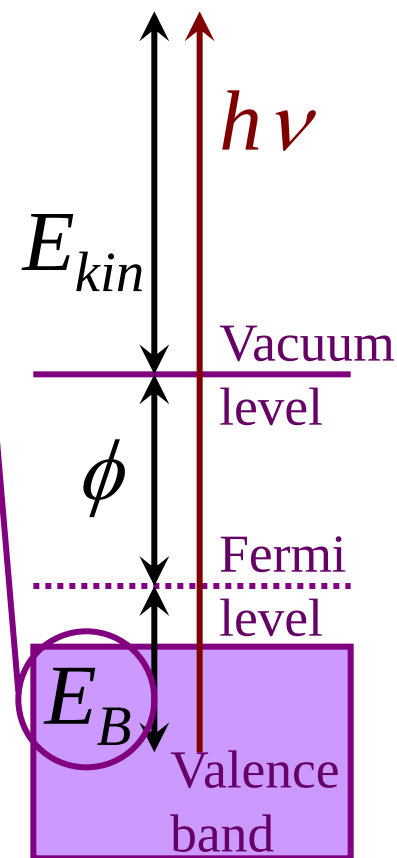
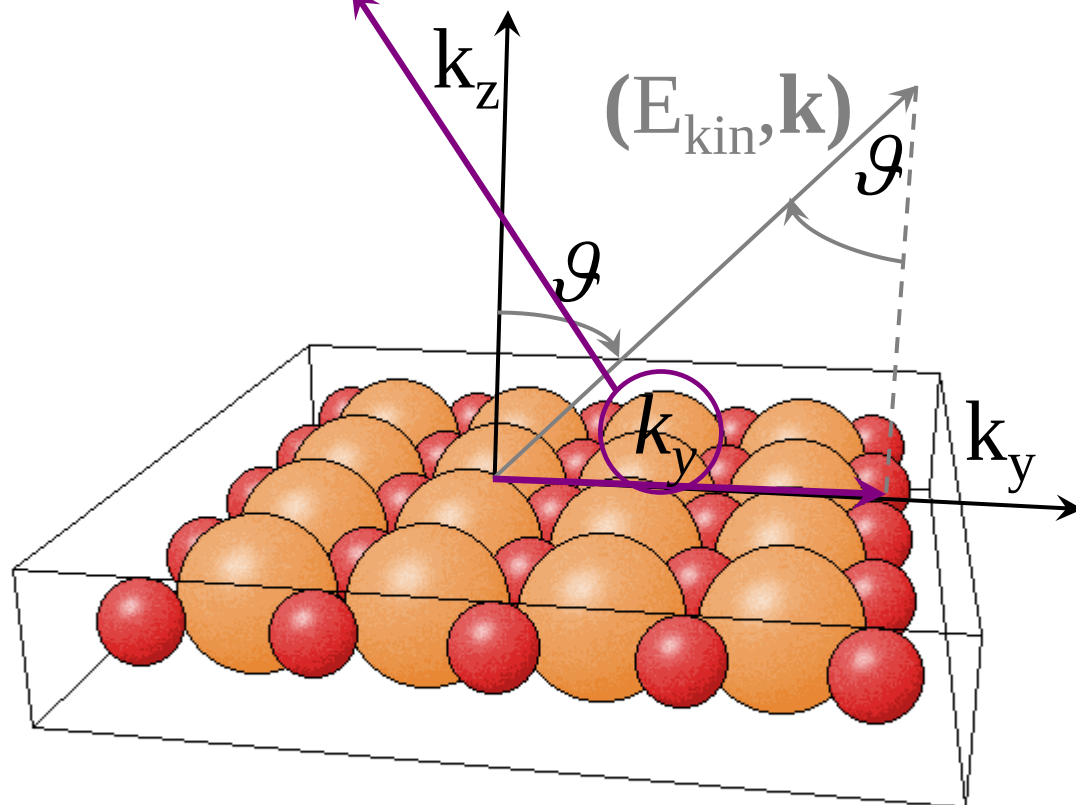
$$k_{i\perp} = \sqrt{\frac{2m}{\hbar^2} (E_{kin} + |E_0|) - G_{\perp}}$$

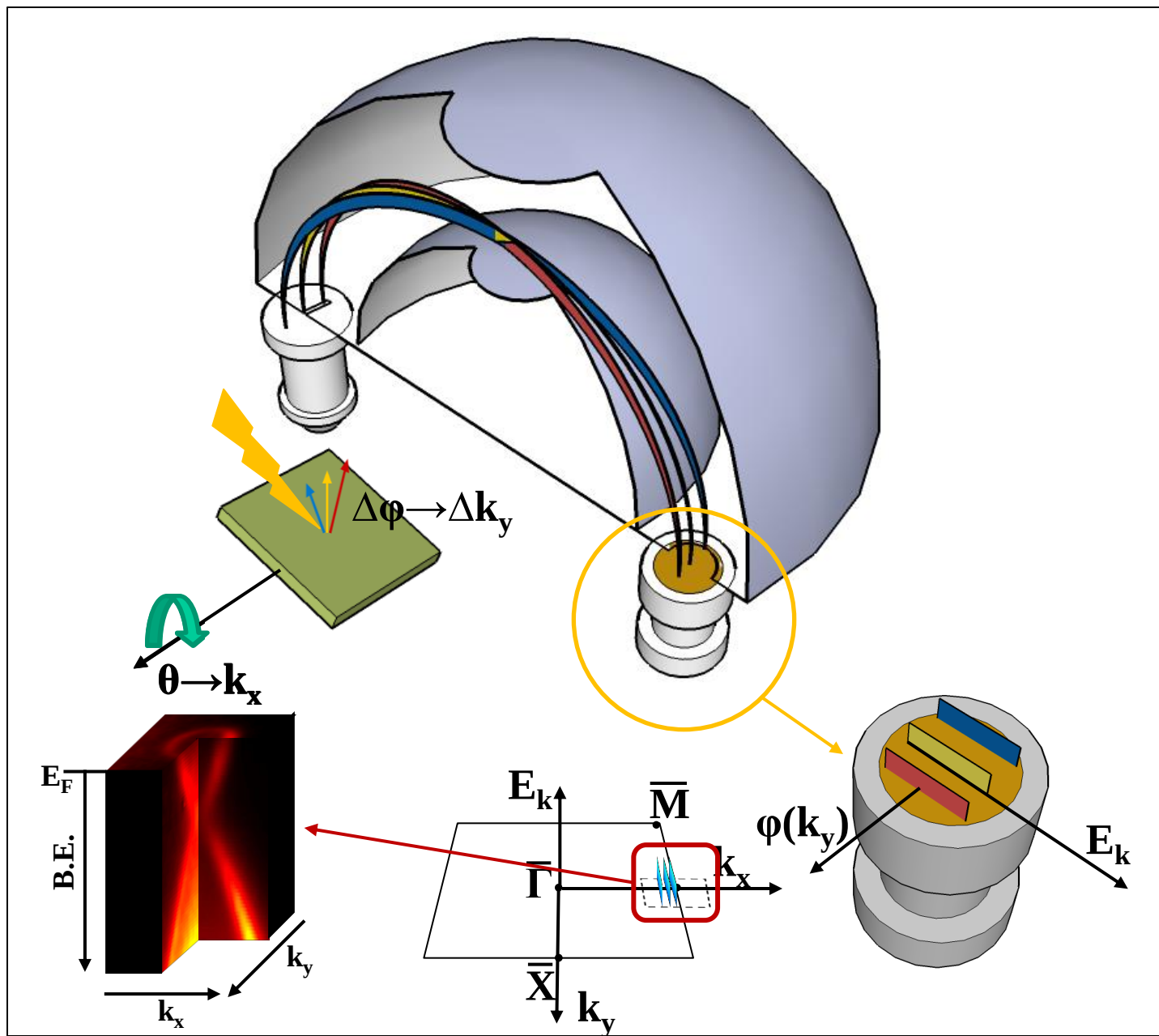




$$k_y = \sqrt{\frac{2m}{\hbar^2} E_{kin}} \cdot \sin(\vartheta)$$

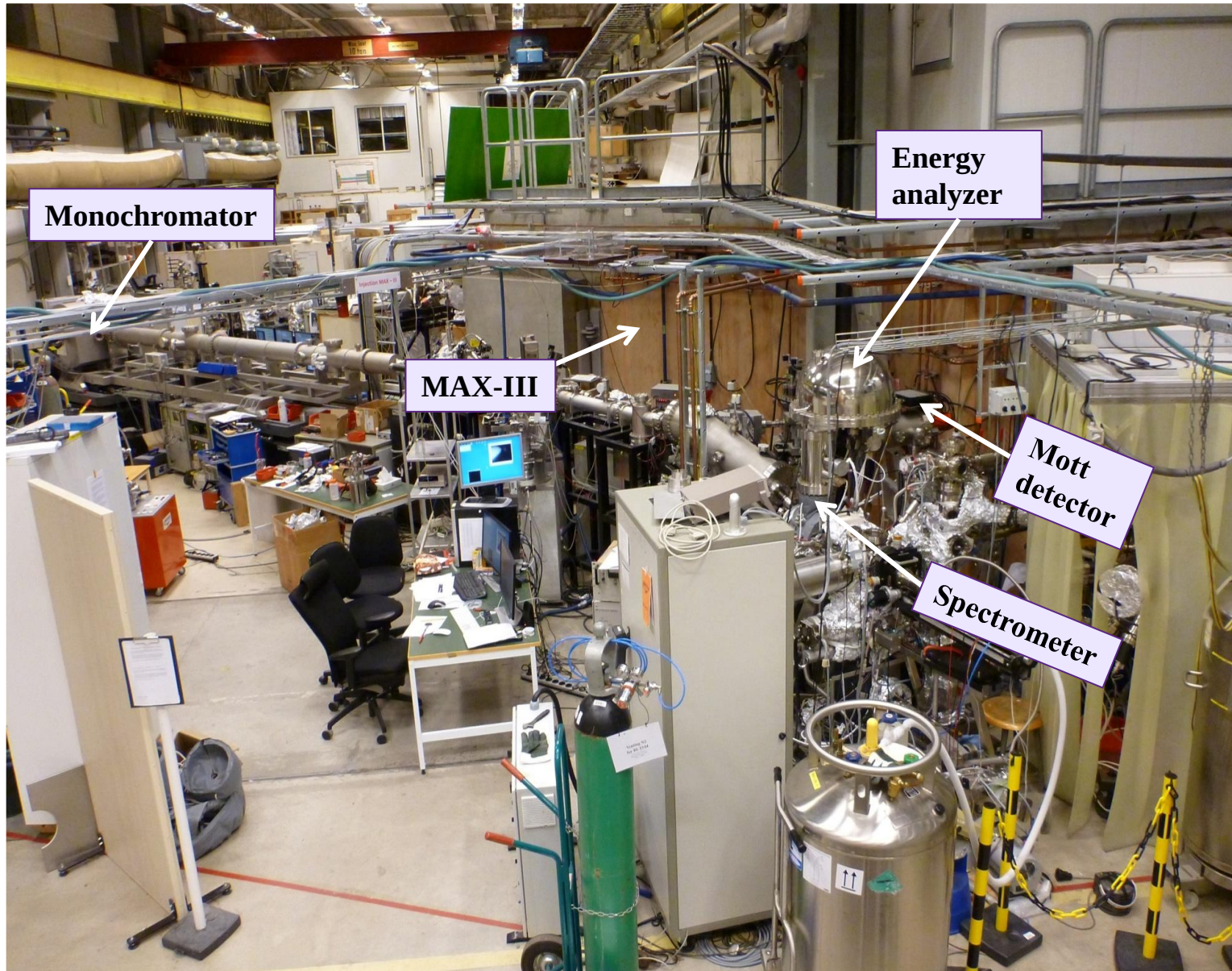
$$E_B = h\nu - (E_{kin} + \phi)$$





Beamline I3

MAX-lab, Lund University, Sweden

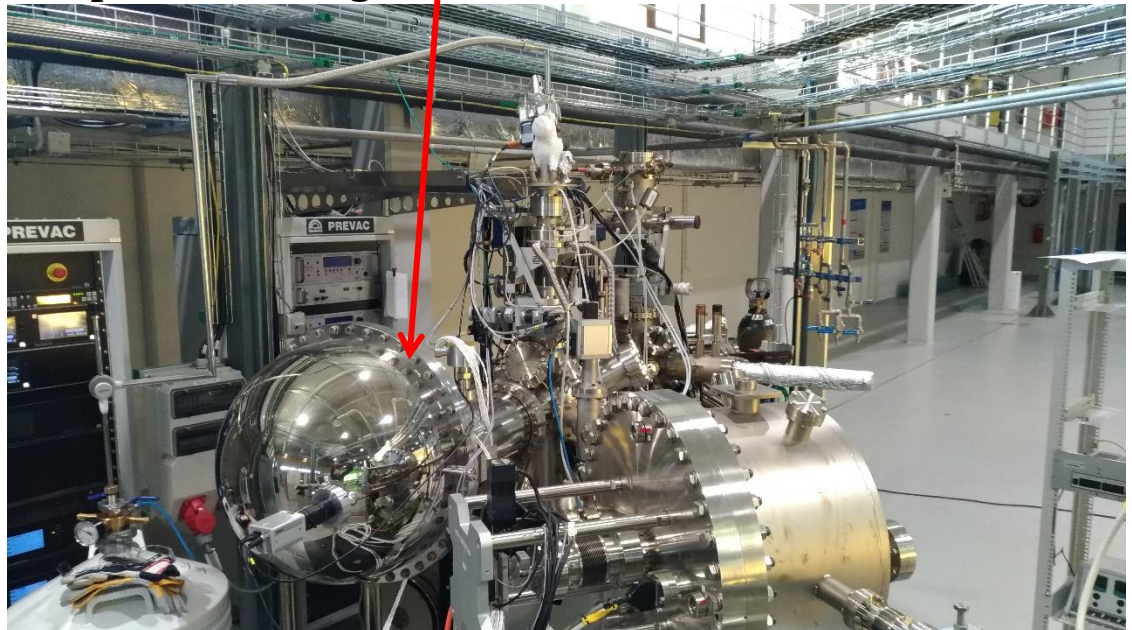


Beamline UARPES

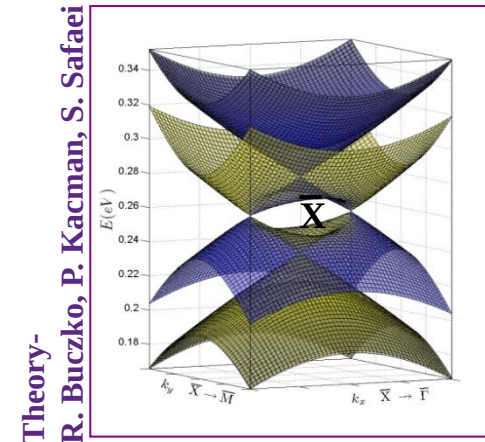
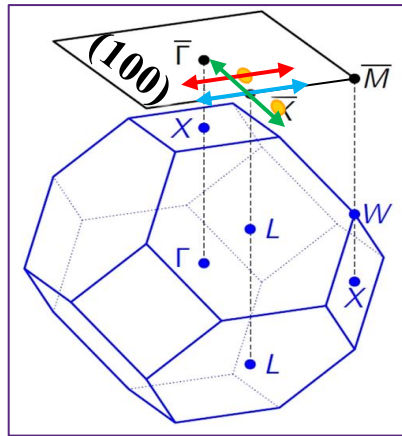
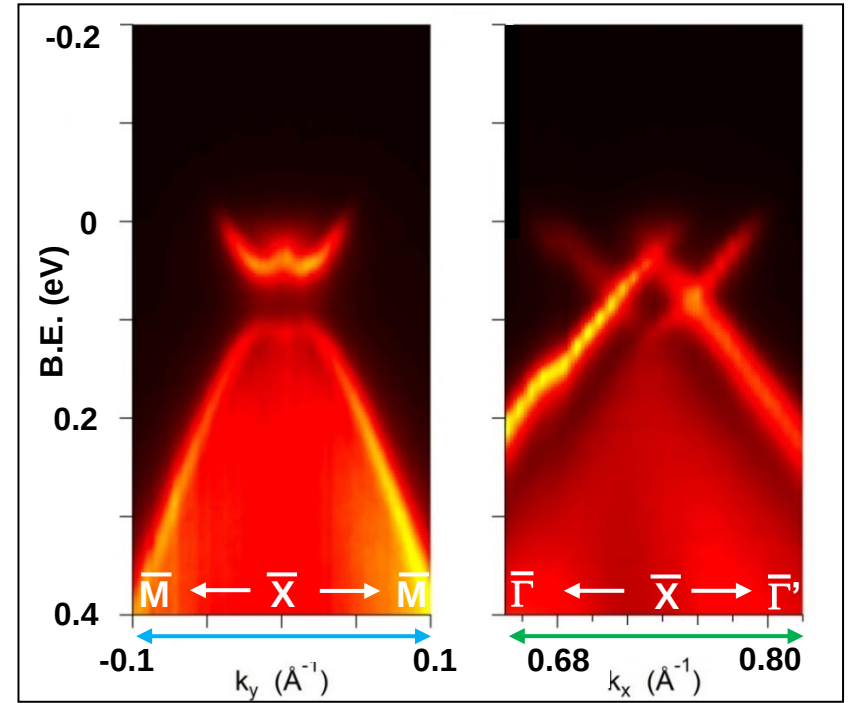
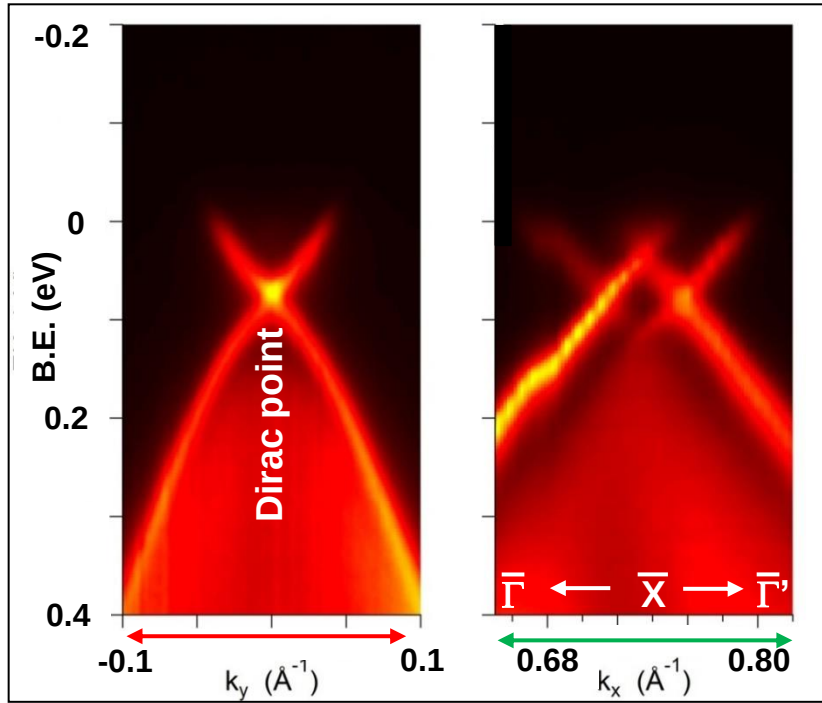
NCSR SOLARIS, Jagiellonian University, Kraków, Poland



- Elliptically polarizing quasiperiodic undulator of APPLE II type
- Monochromator combining normal (NIM) and grazing incidence (PGM) optics (the photon energy range of 8–100 eV)
- SCIENTA OMICRON DA30L photoelectron spectrometer
- The energy and angular resolution: 1.8 meV and 0.1°
- Temperature range 10 – 500 K



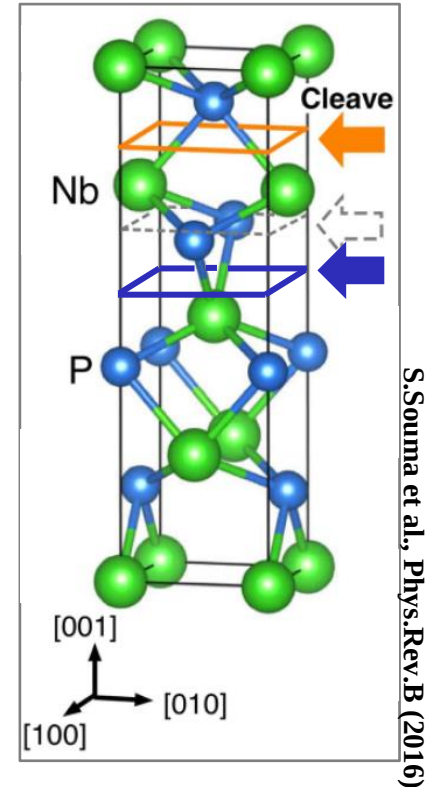
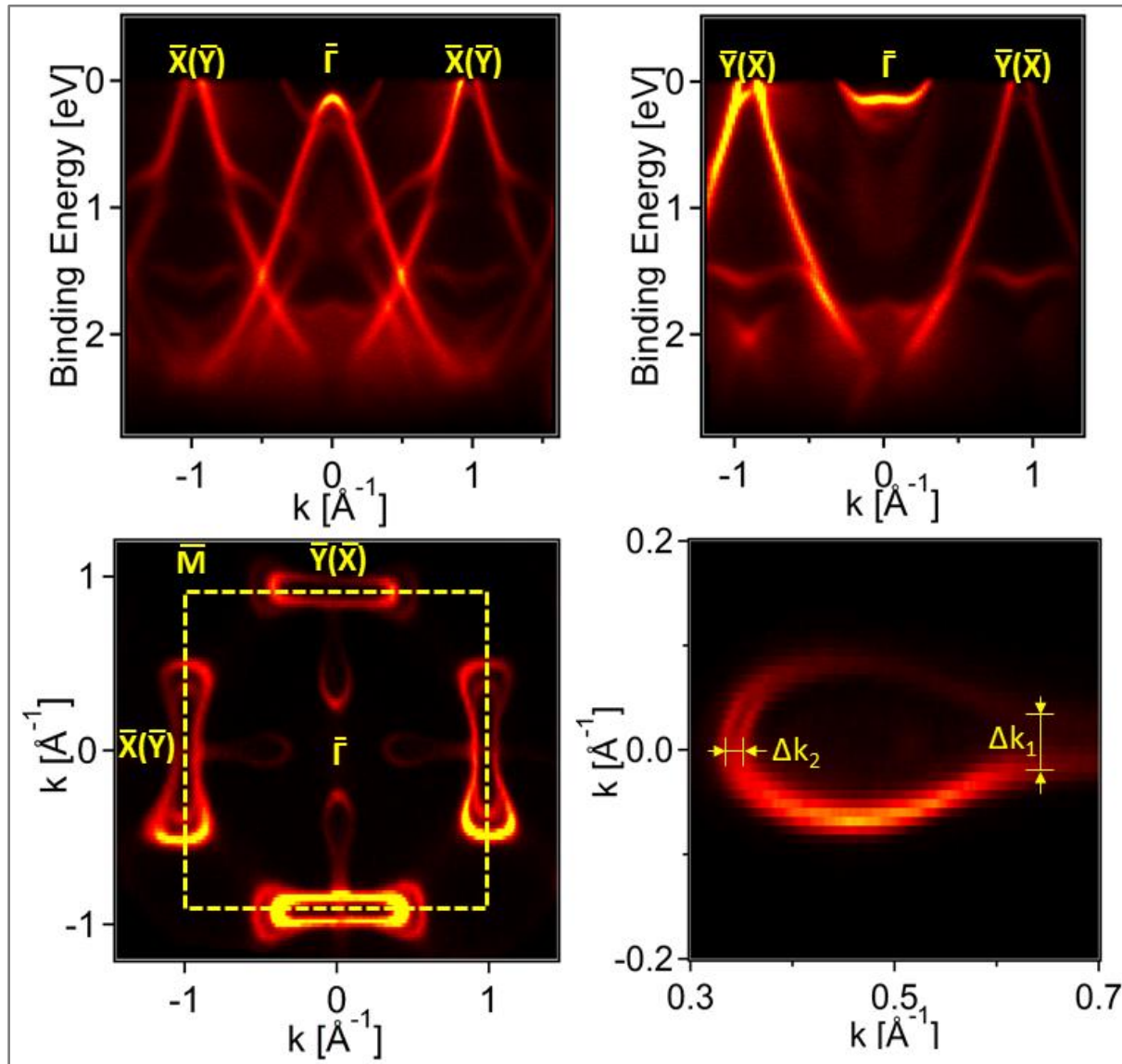
Topological crystalline insulator $\text{Pb}_{0.67}\text{Sn}_{0.33}\text{Se}$, $T=87\text{ K}$, $h\nu=18.5\text{ eV}$



Theory-
R. Buczko, P. Kacman, S. Safaei

P. Dziawa, B. J. Kowalski, K. Dybko, R. Buczko, A. Szczerbakow, M. Szot, E. Łusakowska, T. Balasubramanian, B. M. Wojek, M. H. Berntsen, O. Tjernberg, T. Story, *Nature Materials* **11**, 1023 (2012)

Weyl semimetal NbP



S.Souma et al., Phys.Rev.B (2016)

**ARPES data
for NbP(001)
P-face taken
at UARPES
(SOLARIS)**

Diffraction methods

Surface X-ray diffraction

X-ray total-external-reflection–Bragg diffraction: A structural study of the GaAs-Al interface

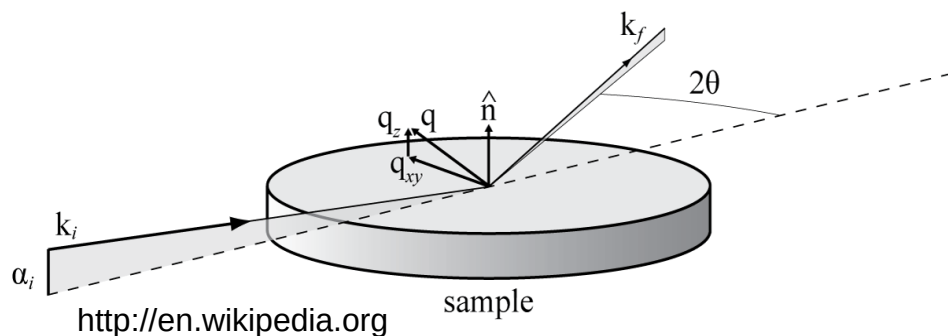
W. C. Marra, P. Eisenberger, and A. Y. Cho

Bell Laboratories, Murray Hill, New Jersey 07974

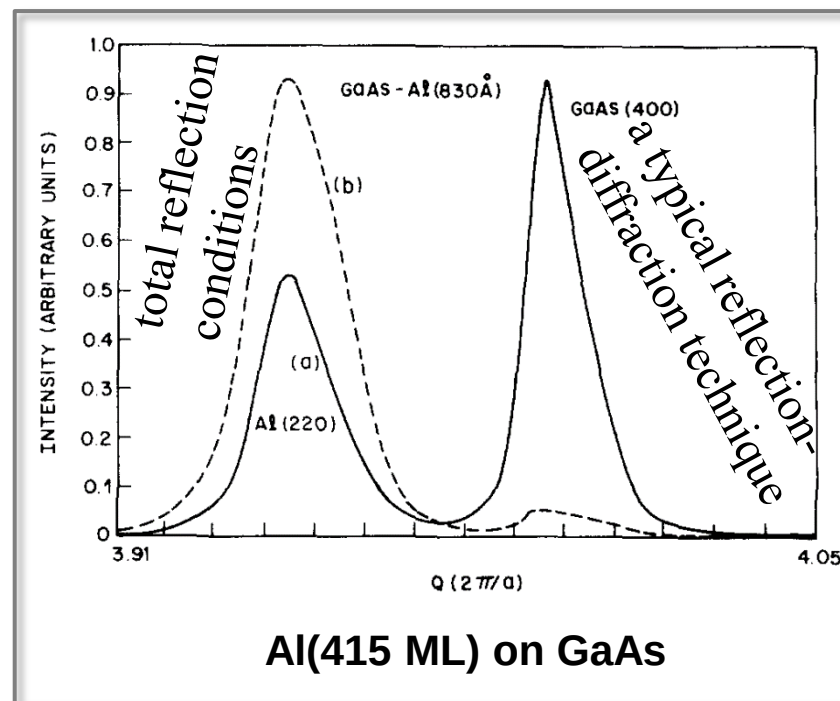
(Received 19 March 1979; accepted for publication 8 June 1979)

A new technique utilizing conventional x-ray diffraction in conjunction with total external reflection has provided a powerful tool for studying ordered interfaces and surface phenomena. It has been used in this work to study the details of the interface region of a molecular beam epitaxially grown Al single crystal on a molecular beam epitaxially grown GaAs single-crystal substrate. A simple model including variations of the lattice parameter and disorder in the interface region is in agreement with these experimental results.

J. Appl. Phys. 50(11), November 1979



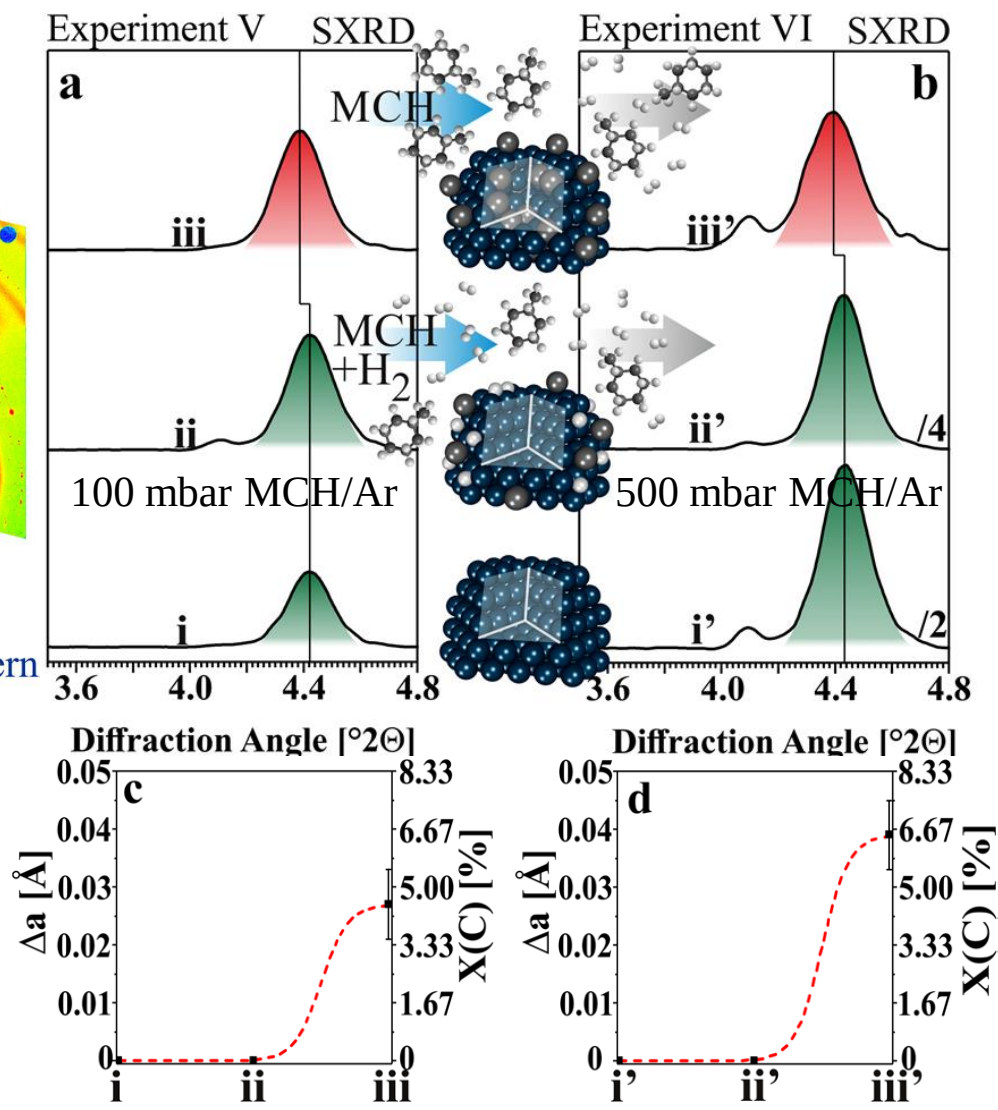
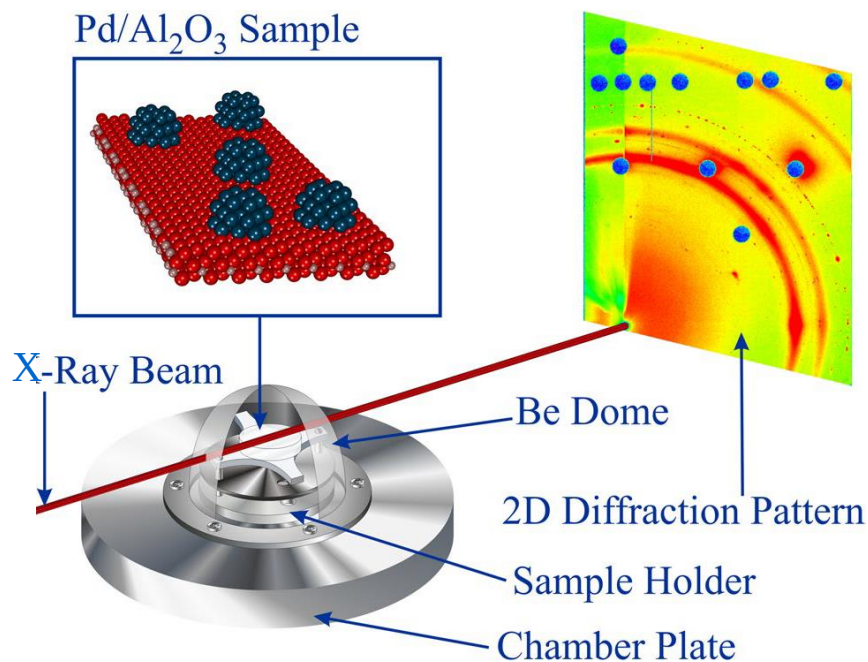
**Needs strong X-ray beam –
usually from a synchrotron!**



Surface X-ray diffraction (cont.)

Dehydrogenation of Liquid Organic Hydrogen Carriers on Supported Pd Model Catalysts: Carbon Incorporation Under Operation Conditions, Ralf Schuster et al., Catalysis Letters **148**, 2901 (2018)

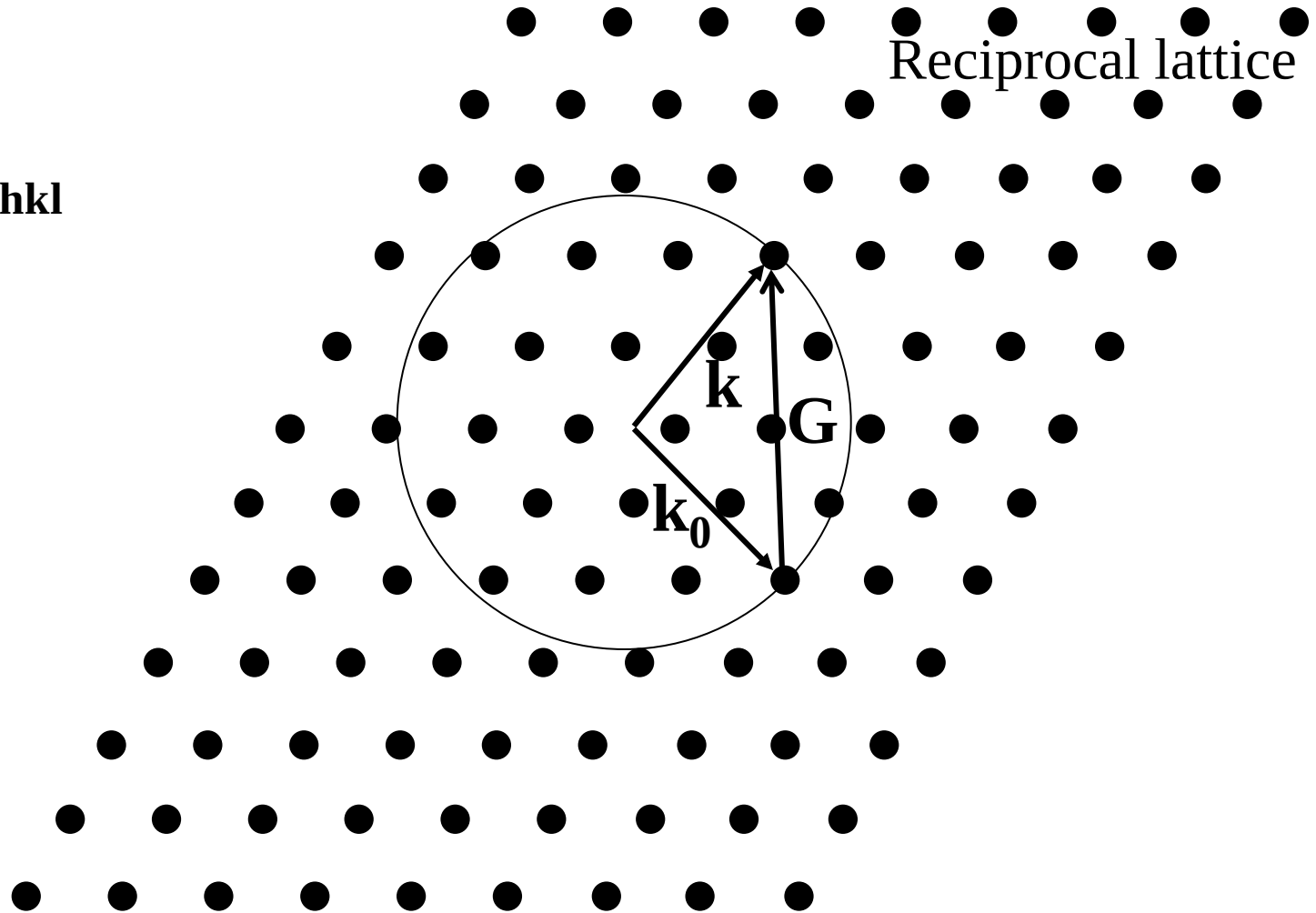
MCH: methylcyclohexane



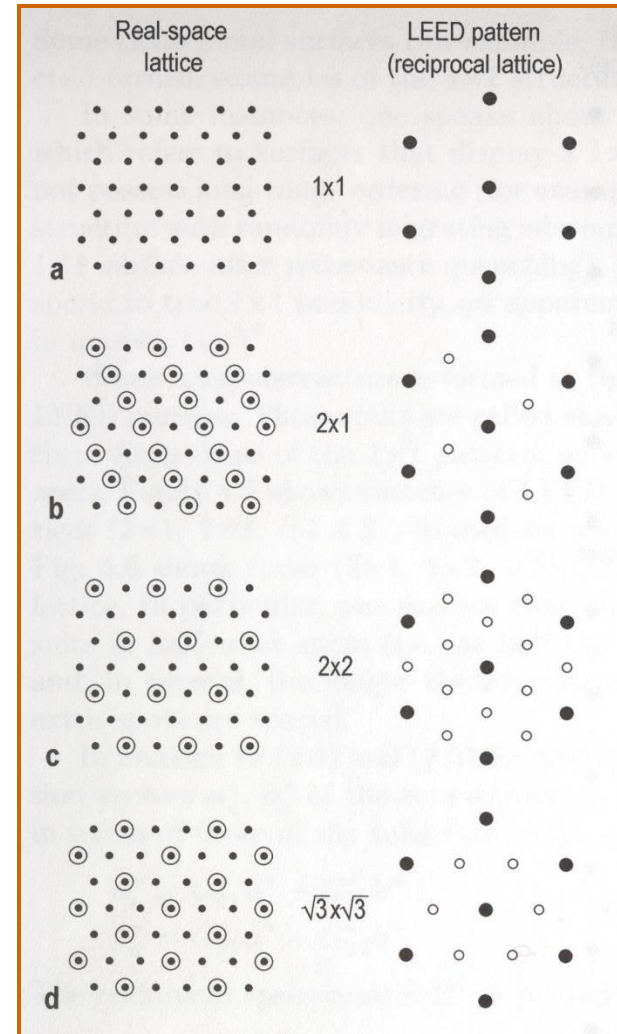
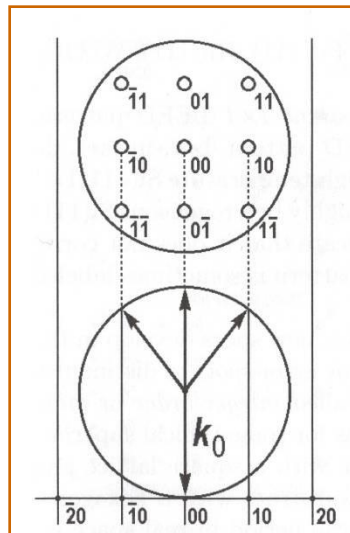
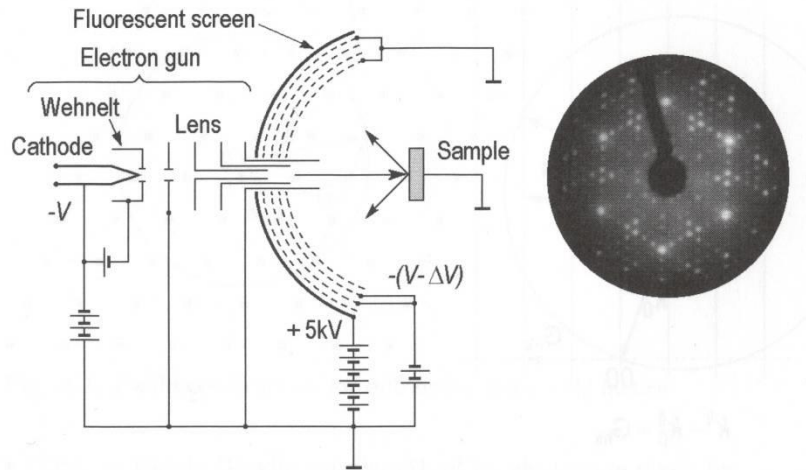
Construction of the Ewald sphere

$$\mathbf{k} - \mathbf{k}_0 = \mathbf{G}_{hkl}$$

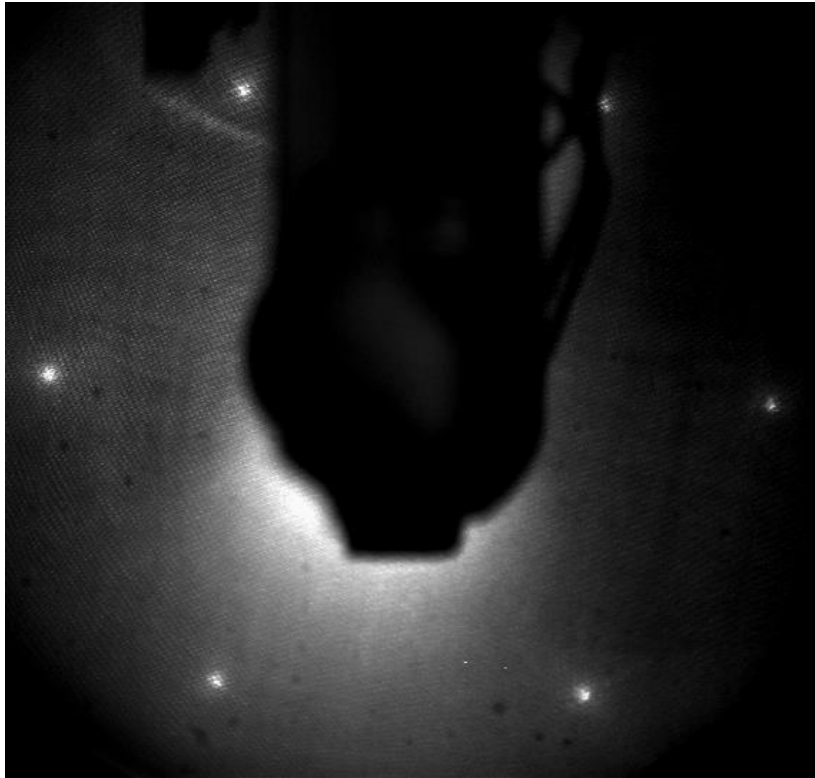
$$|\mathbf{k}| = |\mathbf{k}_0|$$



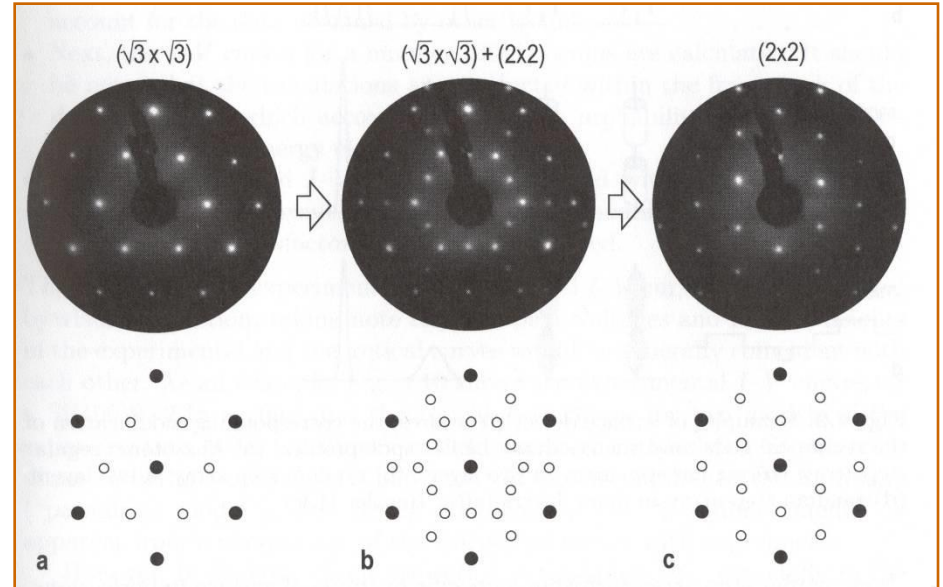
Low-Energy Electron Diffraction (LEED)



Low-Energy Electron Diffraction (LEED) (cont.)

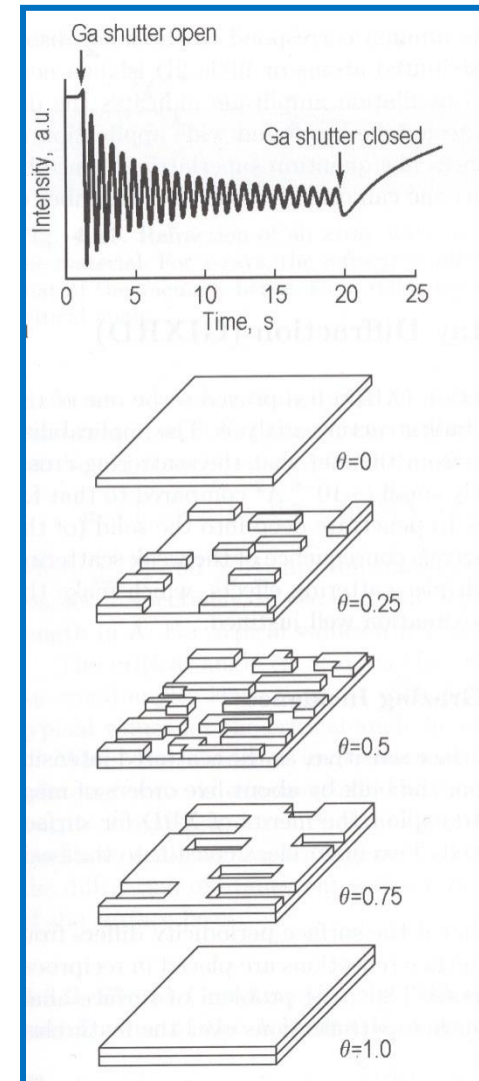
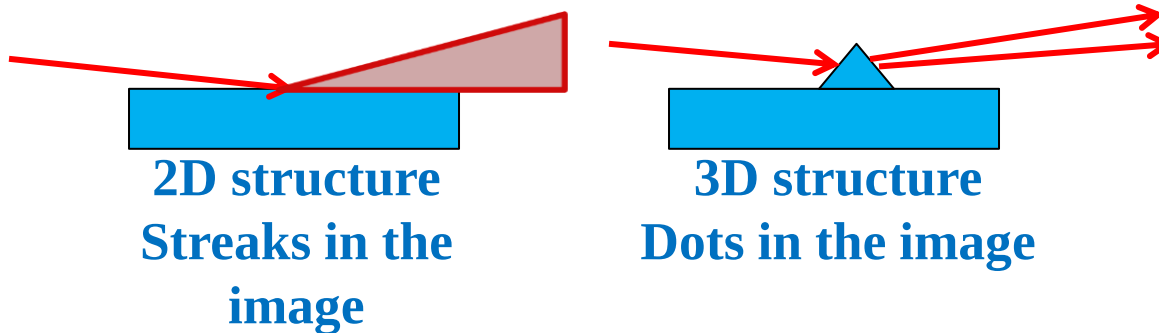
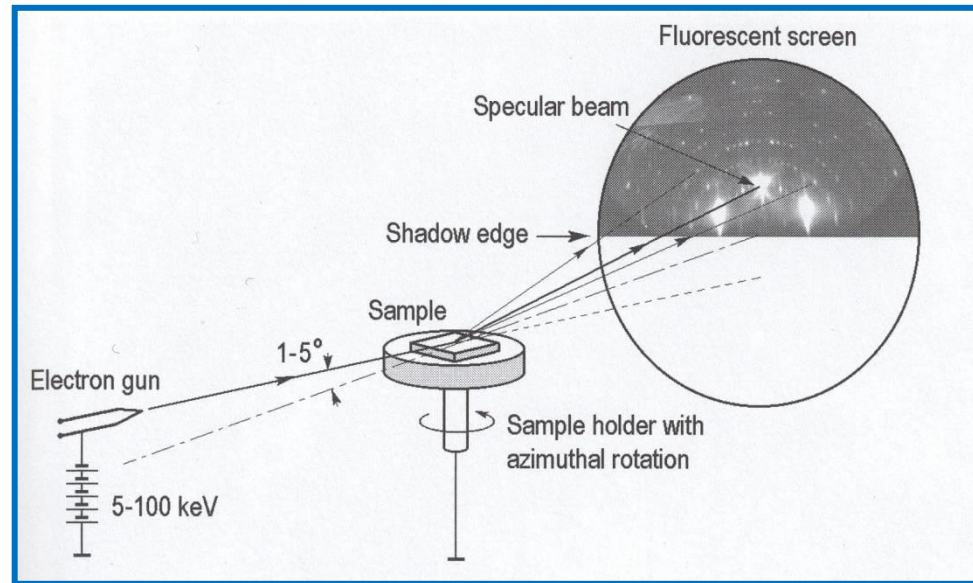


GaN(0001) (1x1)



**In deposition on
Si(111)√3x √ 3-R30⁰**

Reflection High-Energy Electron Diffraction (RHEED)



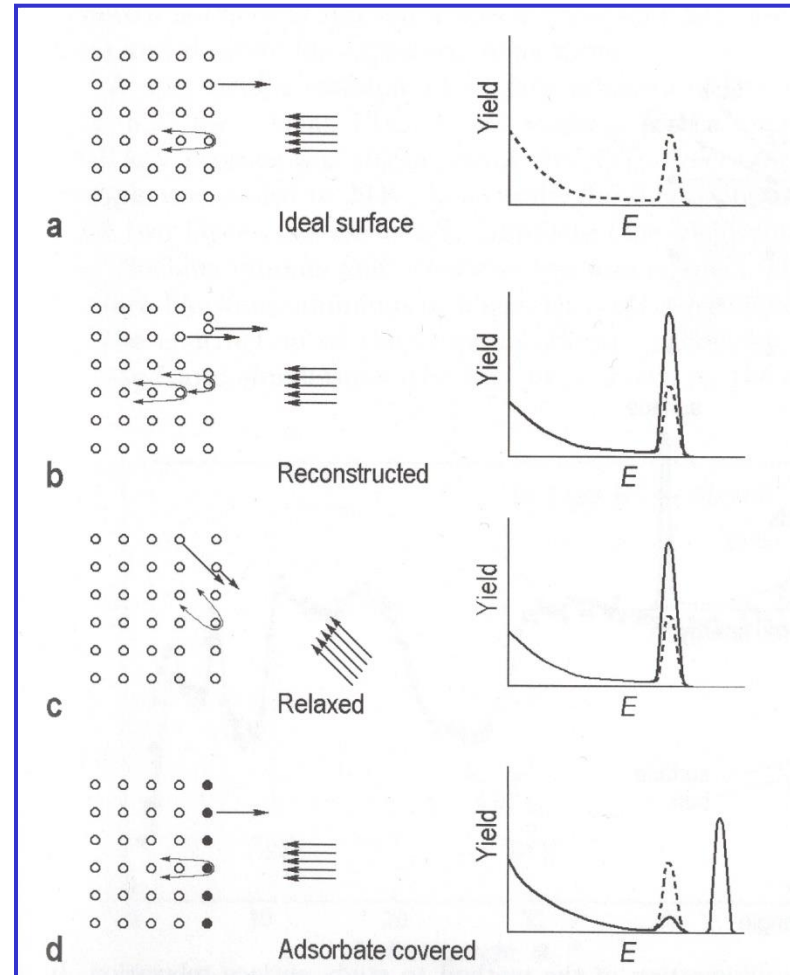
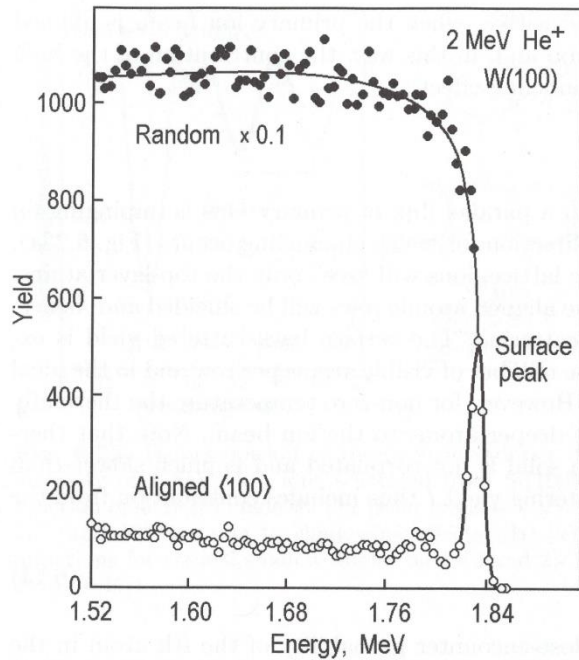
K.Oura et al.
Surface Science. An Introduction

Ion scattering methods

Rutherford Backscattering Spectrometry (RBS)

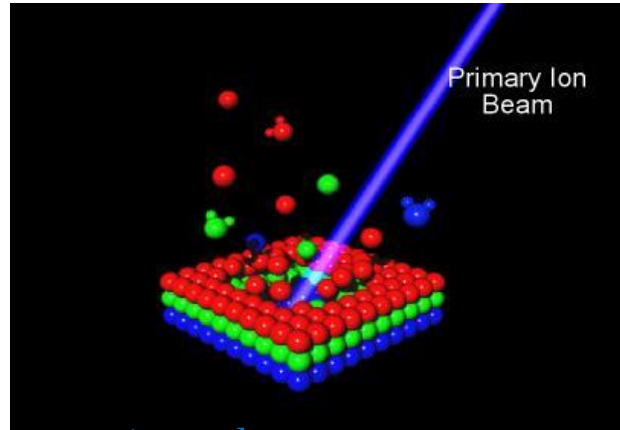
e.g. ^4He
2 MeV

detector

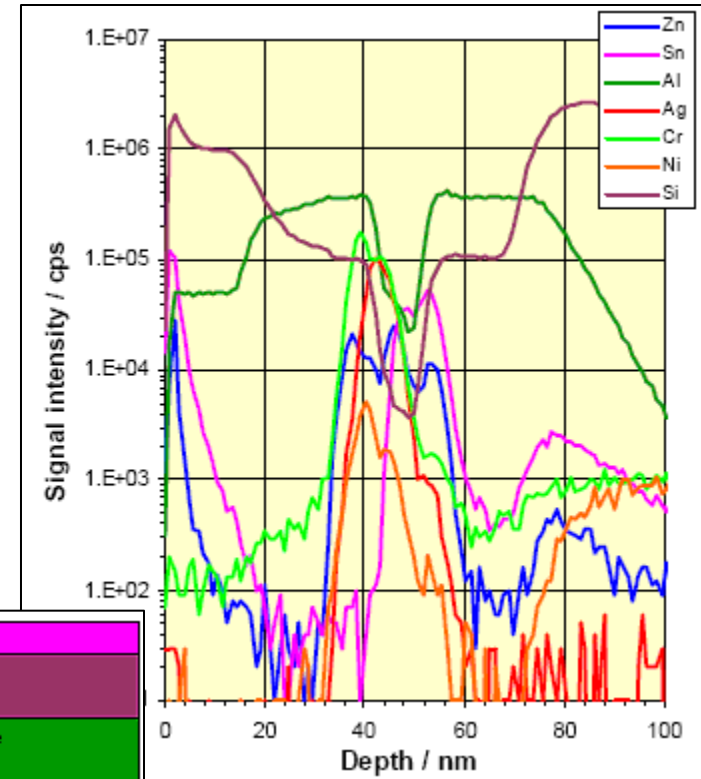


Secondary Ion Mass Spectrometry (SIMS)

e.g. Cs^+ lub Ar^+
1-30 keV



www.ainse.edu.au

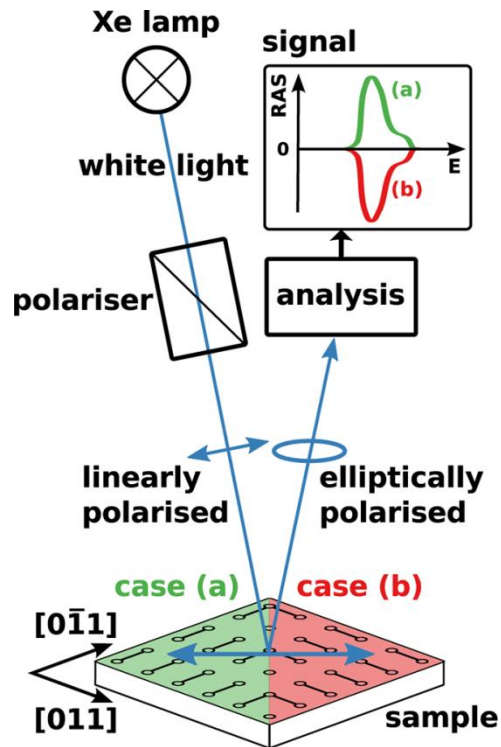


www.azom.com

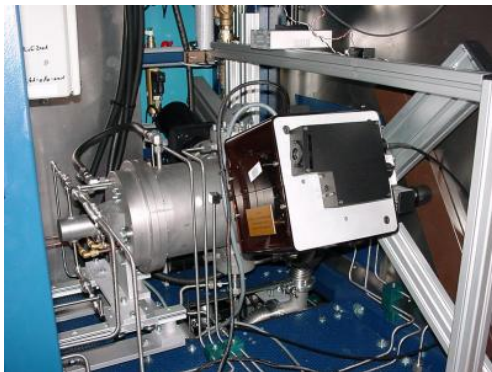
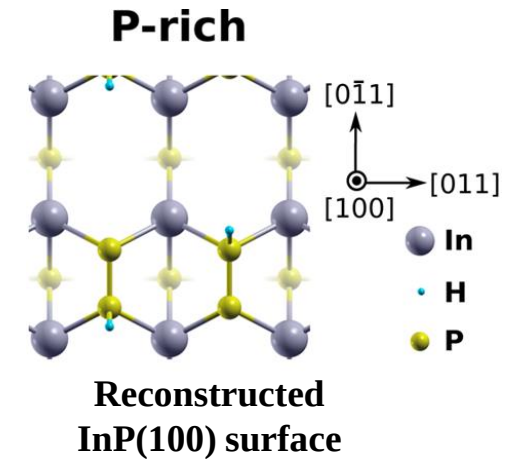
Zinc Tin Oxide
Silicon Nitride
Aluminium Nitride
Zinc Aluminium Oxide
Nickel Chrome Oxide (NiCrOx)
Silver (10 nm)
Zinc Oxide
Zinc Tin Oxide
Aluminium Nitride
Glass Substrate

Optical methods

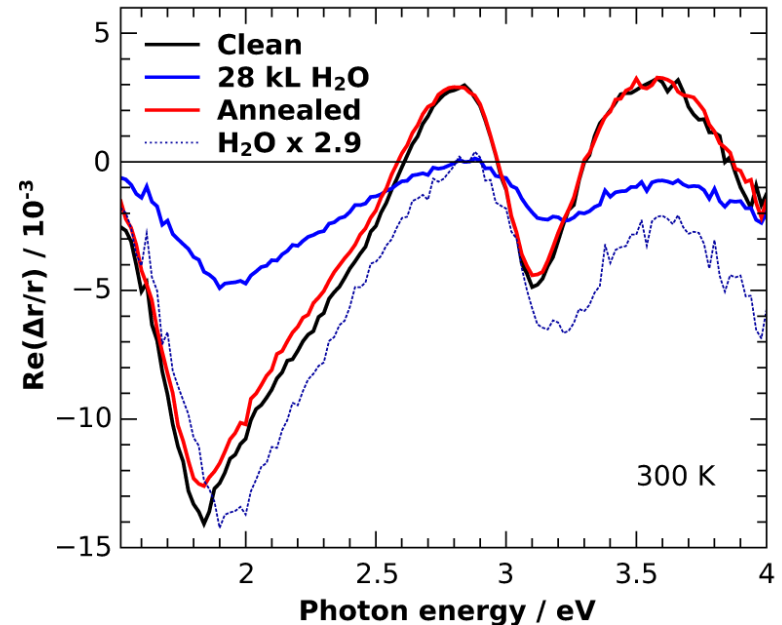
Reflection Anisotropy Spectroscopy (RAS)



$$\frac{\Delta r}{r} = 2 \frac{r_{[0\bar{1}1]} - r_{[011]}}{r_{[0\bar{1}1]} + r_{[011]}}$$



An MOCVD reactor with the RAS system
 Institute of Semiconductor and Solid State
 Physics, University of Linz, Austria



Summary

We can test various surface properties using:

Electron microscopy (SEM)

Scanning probe microscopies (STM, AFM)

Electron spectroscopies (XPS, ARPES, AES)

Diffraction methods (X-ray, LEED, RHEED)

Ion techniques (RBS, SIMS)

Surface-sensitive optical techniques (RAS)

and many others...

Literature:

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De Gruyter 2020

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Cambridge University Press 1988

H. Lüth
Surfaces and Interfaces of Solid Materials
Springer 1995